

Object 16 : Determine the boiling points of ethylene glycol - water solution of different compositions.

Introduction : In case of liquid mixtures, the boiling points depend upon their composition. Consider a liquid mixture of ethylene glycol ($\text{CH}_2\text{OH} - \text{CH}_2\text{OH}$) and water (H_2O) of different composition by volume e.g. 10%, 20%, 30% of ethylene glycol by volume. The boiling points are measured by taking 0.5-1 mL in a fusion glass tube and using a thermometer and Thiele's tube.

Object 17 : Determine the boiling

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M Mosston



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Aids for Teaching Science United States. Office of Education,1963 *Bibliography on Hydrocarbons* J. A. Muckleroy,1962 **Aids for Teaching Science** Ellsworth Scott Obourn,1963 Journal of Chemical Education ,1924
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