#### GUJARAT TECHNOLOGICAL UNIVERSITY B. PHARM. - SEMESTER - IV • EXAMINATION - WINTER 2012

Subject code: 240001 Date: 30/12/2012

Subject Name: Dispensing Pharmacy - I Time: 02.30 pm - 05.30 pm

Total Marks: 80

0.6

05

0.5

0.5

0.5

Instructions:

Q. 6

- 1. Attempt any five questions.
- Make suitable assumptions wherever necessary.
- 3. Figures to the right indicate full marks.
- O.1Define prescription and explain different parts of a typical prescription. (a)
  - Define posology and discuss factors affecting dose calculations. (363)
  - Define draughts, linctuses, drops, syrups and elixirs. (c)
- Q.2Give labeling requirements for inhalations, emulsions, eachets, gargles, dusting 0.6 (a) powders and insufflations.
  - (b) Discuss labeling instructions restricting the use of preparations. 0.5
  - How many tablets each containing 5.85 gr of KMnO, will be required to make 0.5 (c) half a gallon of a 1 in 500 solution?
- If adult dose of a drug is 110 mg. Calculate the dose of child of 6 years using O.3 (a) 06 Young's formula and Dilling's formula
  - Discuss desirable features of containers and closure for dispensing products (b) 0.5 0.5
  - (c) Discuss general dispensing procedure
- How much of 4% ountment must be added to 50 g of an 20% ountment to make an Q.4(a) 0.6 10% ountment 0.5
  - Calculate the gram of solid required to prepare 480 ml of a 1 in 750 ml solution. (b)
  - Discuss solvents used to dissolve or disperse active medicaments. (c)
- Q.5(a) Classify mixtures and discuss diffusible mixture. 06
  - Discuss thickening agents and suspending agents used in indiffusible mixtures. (b) 05 How much taxable alcohol would be in 1 gallon of 45 proof alcohol? (c) 0.5
- - (a) Discuss various tests for the determination of emulsion type. 06 (b) Discuss coalescence phenomenon associated with emulsion stability. 0.5
    - (0)Calculate the amount of NaCl required to make 100 ml of an 2.5 %w/v antibiotic solution isotomic with blood plasma. Freezing point depression of (i) 1 %w/v antibiotic solution is - 0.132, (ii) 1 %w/v NaCl solution is - 0.58 and (iii) blood plasma is - 0.52.
- O. 7 Discuss problems of powders containing deliquescent and efflorescent substances 06 (a) with their remedies.
  - Define displacement value and discuss tablet triturates. (b) 0.5 0.5
  - Send 12 tablet triturates of 1 g each containing 200 mg of ZnO and 300 mg of (c)Bismuth Oxychloride. Displacement value of (i) ZnO is 4.8 and (ii) Bismuth Oxychloride is 4.0.

# **Sample Question Paper For Dispensing Pharmacy**

Mrs. SONY PRADHAN, Mrs.
NITYASHREE MOHAPATRA, , Mr.
DIGAMBAR BISOI, Dr. JITEN MISHRA

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