



### Activity B: Classifying reactions

Get the Gizmo ready:

- Turn off Show summary and Show histograms.



**Introduction:** Chemical equations show how **compounds** and **elements** react with one another. An element is a substance consisting of one kind of atom, such as aluminum (Al) or oxygen gas ( $\text{O}_2$ ). A compound is a substance made of more than one kind of atom, such as water ( $\text{H}_2\text{O}$ ) or table salt ( $\text{NaCl}$ ).

**Question:** How are chemical reactions classified?

1. **Match:** Most chemical reactions can be classified as one of four types. Using the chemical equations in the Gizmo as a guide, match the following definitions to the type of reaction.

B	One reactant is broken down into two or more products.
E	A fuel combines with oxygen to produce carbon dioxide ( $\text{CO}_2$ ) and water ( $\text{H}_2\text{O}$ ).
A	Two or more reactants combine to form one product.
D	Two compounds react to form two different compounds.
C	A compound reacts with an element to form a new compound and a different element.

A. <b>Synthesis</b>
B. <b>Decomposition</b>
C. <b>Single replacement</b>
D. <b>Double replacement</b>
E. <b>Combustion</b>

2. **Practice:** Balance each of the chemical equations below. (Some equations may already be in balance.) In the space to the right, classify the reaction as a *synthesis*, *decomposition*, *single replacement*, *double replacement*, or *combustion*.

A.	<input type="text" value="1"/>	$\text{AgNO}_3 +$	<input type="text" value="1"/>	$\text{KCl} \rightarrow$	<input type="text" value="1"/>	$\text{AgCl} +$	<input type="text" value="1"/>	$\text{KNO}_3$	Two replacements
B.	<input type="text" value="1"/>	$\text{H}_2\text{O} +$	<input type="text" value="1"/>	$\text{SO}_3 \rightarrow$	<input type="text" value="1"/>	$\text{H}_2\text{SO}_4$			Synthesis
C.	<input type="text" value="2"/>	$\text{KI} +$	<input type="text" value="1"/>	$\text{Cl}_2 \rightarrow$	<input type="text" value="2"/>	$\text{KCl} +$	<input type="text" value="1"/>	$\text{I}_2$	Single replacement
D.	<input type="text" value="2"/>	$\text{NaHCO}_3 \rightarrow$	<input type="text" value="1"/>	$\text{Na}_2\text{CO}_3 +$	<input type="text" value="1"/>	$\text{H}_2\text{O} +$	<input type="text" value="1"/>	$\text{CO}_2$	Decomposition
E.	<input type="text" value="1"/>	$\text{Zn} +$	<input type="text" value="2"/>	$\text{HCl} \rightarrow$	<input type="text" value="1"/>	$\text{ZnCl}_2 +$	<input type="text" value="1"/>	$\text{H}_2$	Single replacement

# Balancing Chemical Equations Gizmo Answer Key

**VM Jensen**



## **Balancing Chemical Equations Gizmo Answer Key:**

Science Units for Grades 9-12 Randy L. Bell, Joe Garofalo, 2005 Sample topics include cell division virtual dissection earthquake modeling the Doppler Effect and more Balancing Chemical Equations Worksheet Crispin Collins, 2020-09-12 Struggling with balancing chemical reaction Balancing chemical equations can look intimidating for lot of us The good news is that practice makes perfect Master balancing skill with this workbook packed with hundreds of practice problems This book is for anyone who wants to master the art of balancing chemical reactions First few chapters of this book are step by step explanation of the concepts and other chapters are for practicing problems This book help students develop fluency in balancing chemical equation which provides plenty of practice Methods to solve with the explanation Total of 550 problems to solve with answer key 450 chemical reactions to practice with answer key 100 practice problems that are needed before balancing a chemical reaction with answer key Click the Buy now button to take advantage of this book to help yourself in mastering balancing skill *Balancing Chemical Equations Workbook* Kraaya Publishing, 2021-02-21 Chemical Reactions to Balance Workbook This chemistry balancing equations practice workbook contains 250 non balanced chemical equations Begin with 2 terms problems Work your way up to 6 terms problems This is the perfect workbook to increase chemistry balancing skills for beginners Table of contents How To Balance A Chemical Equation Chemical Equations To Balance Correct Answers Book features Non repetitive equations Include all reactions types synthesis combustion decomposition Use it now and develop instant recall of balancing equations Enjoy the challenge **Balancing Chemical Equations** Rumi Michael Leigh, 2023-05-22 Do you find yourself struggling to balance chemical equations Are you searching for a comprehensive guide that will help you overcome the challenges of this fundamental skill Look no further Balancing Chemical Equations things you should know questions and answers is here to transform your understanding and proficiency in this crucial aspect of chemistry This book is a practical and engaging resource designed to provide learners of all levels with a solid foundation in balancing chemical equations Whether you re a student a self learner or a passionate science enthusiast this guide will equip you with the essential techniques and strategies required to tackle chemical equations with confidence and precision By actively participating in the exercises you ll develop a deep understanding of the principles and enhance your problem solving abilities Whether you re preparing for an exam aiming to excel in your chemistry coursework or simply eager to master this crucial skill Balancing Chemical Equations things you should know questions and answers is your ultimate companion **Balancing Chemical Equations Worksheets (Over 200 Reactions to Balance)** Chris McMullen, 2016-01-12 Master the art of balancing chemical reactions through examples and practice 10 examples are fully solved step by step with explanations to serve as a guide Over 200 chemical equations provide ample practice Exercises start out easy and grow progressively more challenging and involved Answers to every problem are tabulated at the back of the book A chapter of pre balancing exercises helps develop essential counting skills Opening chapter reviews pertinent concepts

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*Balancing Chemical Equations* Alan Trevor Neuff, 1969

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**Balancing Act! How to Balance a Chemical Equation and the Law of Conservation of Mass | Grade 6-8 Physical Science** Baby Professor,2024-04-15 Explore the intriguing principles of chemical balance with this essential guide perfect for middle school educators homeschooling parents and librarians This book unravels the Law of Conservation of Mass and demystifies the process of balancing chemical equations underscoring their significance in the US STEM curriculum Gain insights into the immutable nature of mass in chemical reactions and master the art of equation balancing to foster a deep understanding of chemical transformations a must have resource for promoting scientific literacy in young learners

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