solvent

increases

solute

Overview Solutions

Directions: Use the words listed below to complete the paragraphs. Some words will be used more than once.

decreases electrolyte	temperature nonelectrolyte	nonpolar higher	polar
2	is the substance	being dissolved. For	solids dissolving in
liquids, the speed of dissolving 3.		with increased surface	
area, with increased	temperature, and with	stirring. For gases in	liquids, the speed o
solubility 4.	ubility 4 with increased temperature but 5.		
	with increased pr	essure.	
In general, polar	solutes dissolve in pol	ar solvents and do no	t dissolve in
6	solvents. Usually, nonpolar solvents dissolve nonpolar		
solutes and do not d	issolve 7.	solutes.	
Solubility is usu	ally expressed as the ma	aximum number of g	rams of
8	that will dissolve in 100 grams of solvent at a certain		
9			
A substance that	forms charged ions in	solution and can con	duct electricity is
called a(n) 10.	A	n) 11	does not
ionize in water and	cannot conduct electric	ity. The boiling point	of a water solution
of any substance is 1:	2 than the boiling point of water.		
The freezing point of	a water solution of any	substance is 13	
than the freezing poi	nt of water.		

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