

GUJARAT TECHNOLOGICAL UNIVERSITY**B. PHARM. - SEMESTER – IV • EXAMINATION – WINTER 2012****Subject code: 240001****Date: 30/12/2012****Subject Name: Dispensing Pharmacy – I****Time: 02.30 pm – 05.30 pm****Total Marks: 80****Instructions:**

1. Attempt any five questions.
2. Make suitable assumptions wherever necessary.
3. Figures to the right indicate full marks.

Q.1	(a)	Define prescription and explain different parts of a typical prescription.	06
	(b)	Define posology and discuss factors affecting dose calculations.	05
	(c)	Define draughts, linctuses, drops, syrups and elixirs.	05
Q.2	(a)	Give labeling requirements for inhalations, emulsions, cachets, gargles, dusting powders and insufflations.	06
	(b)	Discuss labeling instructions restricting the use of preparations.	05
	(c)	How many tablets each containing 5.85 gr of KMnO_4 will be required to make half a gallon of a 1 in 500 solution?	05
Q.3	(a)	If adult dose of a drug is 110 mg. Calculate the dose of child of 6 years using Young's formula and Dilling's formula.	06
	(b)	Discuss desirable features of containers and closure for dispensing products.	05
	(c)	Discuss general dispensing procedure.	05
Q.4	(a)	How much of 4% ointment must be added to 50 g of an 20% ointment to make an 10% ointment.	06
	(b)	Calculate the gram of solid required to prepare 480 ml of a 1 in 750 ml solution.	05
	(c)	Discuss solvents used to dissolve or disperse active medicaments.	05
Q.5	(a)	Classify mixtures and discuss diffusible mixture.	06
	(b)	Discuss thickening agents and suspending agents used in indiffusible mixtures.	05
	(c)	How much taxable alcohol would be in 1 gallon of 45 proof alcohol?	05
Q. 6	(a)	Discuss various tests for the determination of emulsion type.	06
	(b)	Discuss coalescence phenomenon associated with emulsion stability.	05
	(c)	Calculate the amount of NaCl required to make 100 ml of an 2.5 %w/v antibiotic solution isotonic with blood plasma. Freezing point depression of (i) 1 %w/v antibiotic solution is - 0.132, (ii) 1 %w/v NaCl solution is - 0.58 and (iii) blood plasma is - 0.52.	05
Q. 7	(a)	Discuss problems of powders containing deliquescent and efflorescent substances with their remedies.	06
	(b)	Define displacement value and discuss tablet triturates.	05
	(c)	Send 12 tablet triturates of 1 g each containing 200 mg of ZnO and 300 mg of Bismuth Oxychloride. Displacement value of (i) ZnO is 4.8 and (ii) Bismuth Oxychloride is 4.0.	05

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Rachel Sandford



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