

Guru:	Anggayudha	Kelas/Smt:	XII/1	Tgl ajar:	16/08/2016	Jam ajar:	5-6
Mata Pelajaran:	Kimia	Alokasi waktu: 2x40 menit					
Topik bahasan utama: Sifat-sifat periodic unsure							
Tujuan Pembelajaran:							
<ul style="list-style-type: none">- Siswa memahami keterkaitan antara konfigurasi electron pada atom dengan ukuran atom- Siswa memahami keterkaitan antara ukuran atom dengan kemampuan atom untuk membentuk ion positif dan negative- Siswa mampu membandingkan unsure dengan energi ionisasi lebih rendah dan lebih tinggi- Siswa mampu membandingkan unsure dengan afinitas electron lebih rendah dan lebih tinggi- Siswa mampu membandingkan unsure dengan elektronegativitas lebih rendah dan lebih tinggi							
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Media/referensi yang dibutuhkan:	<ul style="list-style-type: none">- Molymod- Magnet batang 6 buah- Kancing 1 paket- Selotip						

Alur/Metode Pengajaran

Pembukaan:

- Ketik disini

Inti:

- Pembahasan tentang ukuran atom
- Pembahasan tentang kekuatan tarik-menarik positif-negatif dan tolak menolak positif-positif dan negative-negatif
- Simulasi energi ionisasi
- Penjelasan energi ionisasi lebih rendah dan lebih tinggi
- Penjelasan afinitas electron
- Simulasi afinitas electron
- Penjelasan keelektronegatifan

Penutup:

- Dengan memahami hal ini, kita semakin mudah memahami mana unsure yang bereaksi satu sama lain

Inspirasi Yang dibagikan:

Ilmu kimia = ilmu yang membuat kita semakin dekat dengan pencipta. Semakin kita memahami kimia, semakin kita memahami betapa rumit proses penciptaan. Padahal Allah selalu melakukan penciptaan itu SETIAP SAAT. Betapa agung nya Kuasa Allah.

Menyetujui

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CL Gary



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web moons also called satellites come in many shapes sizes and types they are generally solid bodies and few have atmospheres most of the planetary moons probably formed from the discs of gas and dust circulating around planets in the early solar system some moons are large enough for their gravity to cause

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