

**PO ON COMMERCIAL ASSOCIATION WONG SIU CHING SECONDARY SCHOOL**  
**2015 – 2016 First Term Examination**  
**Chemistry**

Form 3

Time: 9:45 – 10:30 a.m.

Full Mark: 100

Date: 14 January, 2016

No. of Pages: 2

Setter: Lau Hoi Yi

Answer ALL questions. Put all your answers in the answer sheets provided.

**Section A: Multiple Choices (39%)**

1. What is the meaning of the following hazard warning label?

- A Flammable                      B Carcinogenic  
 C Oxidizing                      D Harmful



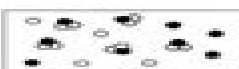
2. Which of the above diagrams best represents the arrangement of particles of a mixture of chlorine and argon at room temperature and pressure?



A



B



C



D

3. The table shows the melting points and boiling points of four substances at 1 atm pressure.

Substance	Melting point (°C)	Boiling point (°C)
W	-60	-5
X	-189	-55
Y	-40	60
Z	5	58

Which substance exists as a liquid at -50°C and room pressure?

- A W                      B X                      C Y                      D Z
- A Bromine                      B Magnesium                      C Nitrogen                      D Sulphur

5. Which of the following elements is a metalloid?

- A S                      B Ne                      C Br                      D Si

6. Atoms are electrically neutral because

- A neutrons are neutral.                      B they have the same number of neutrons and protons.  
 C all subatomic particles are neutral.                      D they have the same number of protons and electrons.

7. Which of the following combinations concerning an atom  $^{88}_{37}\text{X}$  is correct?

	Number of electrons	Number of neutrons	Number of protons
A	48	48	37
B	37	37	48
C	37	48	37
D	48	37	48

8. Which of the following species contains the same number of neutrons as  $^{27}_{13}\text{Al}$ ?

- A  $^{25}_{12}\text{Mg}$                       B  $^{20}_{10}\text{Ne}$                       C  $^{32}_{16}\text{S}$                       D  $^{28}_{14}\text{Si}$

9. Which of the following is most likely to represent an isotope of  $^{52}_{24}\text{X}$ ?

- A  $^{52}_{23}\text{X}$                       B  $^{52}_{24}\text{X}$                       C  $^{53}_{24}\text{X}$                       D  $^{72}_{24}\text{X}$

10. The following is an electron diagram of an atom of element Y:



(Only electrons in the outermost shell are shown.)

The atomic number of Y could be

- A 5                      B 7                      C 11                      D 13

11. Which of the following pairs of elements belong to the same period?

- A Boron and fluorine                      B Beryllium and sulphur                      C Magnesium and neon                      D Potassium and helium

# F3 Chemistry Paper

**Rachel Sandford**



### **F3 Chemistry Paper:**

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