Experimental and Mathematical Modeling of the Effect of Pollution Constitution on DC Flashover of High Voltage Insulator

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Abstract: This paper is aimed at the study of the influence of salts and salts mixtures on dc flashover discharge of high voltage polluted insulators. It's shown that the type of salt and salts mixtures don't have a significant influence on the critical voltages at both polarities. The critical voltages computed using the dynamic equations of dc flashover discharge characteristic constants (n and N) of polluted insulators we established elsewhere are found in a good accordance with the experimental ones. It also appears from this study that the most important components of the discharge column are water vapour and air and the transfer of metal ions has not a significant influence on flashover voltage.

Index Terms — Pollution, Insulators, Flashover, Modeling, Discharge characteristic constants, Constitution of pollution layer, Salts, Salts Mixtures.

1 INTRODUCTION

THE Flashover of polluted high-voltage insulators is a common problem in power electrical networks. The accumulation of pollution layer on the insulator surface is the first stage of flashover process. The constitution of this pollution differs from one site to another. However, the main elements that are found are minerals and insoluble materials [1-7]. Numerous studies have been conducted to determine the influence of pollution constitution on ac and dc flashover characteristics [8-12]. Ghosh and Chatterjee [12] showed that the constituents of pollution deposits represent a crucial parameter that can affect the acflashover characteristics. They conclude that the discharge electrical gradient is affected by the nature of electrolyte. This influence is reflected by the variation of values of the discharge characteristic constant N and n with the type of salt within the pollution deposit (Table 1). The critical current versus the linear resistance of pollution with NaCl is greater than that with other salts [12]. As concerns the variation of critical voltage versus the linear resistance of pollution, these authors don't give any indication. By measuring the quantities of metallic ions before and after flashover, they found that these quantities increase; and they conclude that there is a transport phenomenon of metallic ions during flashover process. Matsumoto et al. [13], show that in the first step of discharge propagation, the discharge column contains traces of No and Na. They conclude

that the presence of metallic atom has an influence on the temperature of the discharge comparing to a free air discharge. In electrochemistry field, many authors investigated the interaction of an electrical discharge with a water surface or volume [14 - 17]. It was shown that the most components of chemical species in discharge column are water vapour and air and there is a transfer of a few part of metallic ions from the liquid. The presence of metallic ions is explained with the sputtering phenomena from the liquid to plasma. The amounts of metallic ions depends on their concentration within the liquid.

The aim of this paper is to study the experimental effect of different salts and salts mixtures on dc critical flashover voltage V_{on}. The experimental values of V_{on} are compared to those deduced from the relationships we established elsewhere [2].

Table 1: Values of flashover characteristic constants n and N according to Ghosh [12]

Electrolyte	N	/9
NaCi	360	0.59
CaCle	461	0.42
FeOL	270	0.66
CuSO ₄	450	0.49

2 EXPERIMENTAL SETUP

The experimental device we used is the same as that presented in a previous work [11, 18]. It consists of a

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- 1. Understanding the eBook Experimental Study And Mathematical Modeling Of Flashover
 - The Rise of Digital Reading Experimental Study And Mathematical Modeling Of Flashover
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Experimental Study And Mathematical Modeling Of Flashover
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - \circ Features to Look for in an Experimental Study And Mathematical Modeling Of Flashover
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Experimental Study And Mathematical Modeling Of Flashover
 - Personalized Recommendations
 - Experimental Study And Mathematical Modeling Of Flashover User Reviews and Ratings
 - Experimental Study And Mathematical Modeling Of Flashover and Bestseller Lists
- 5. Accessing Experimental Study And Mathematical Modeling Of Flashover Free and Paid eBooks
 - Experimental Study And Mathematical Modeling Of Flashover Public Domain eBooks
 - Experimental Study And Mathematical Modeling Of Flashover eBook Subscription Services

- Experimental Study And Mathematical Modeling Of Flashover Budget-Friendly Options
- 6. Navigating Experimental Study And Mathematical Modeling Of Flashover eBook Formats
 - ePub, PDF, MOBI, and More
 - Experimental Study And Mathematical Modeling Of Flashover Compatibility with Devices
 - Experimental Study And Mathematical Modeling Of Flashover Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Experimental Study And Mathematical Modeling Of Flashover
 - Highlighting and Note-Taking Experimental Study And Mathematical Modeling Of Flashover
 - Interactive Elements Experimental Study And Mathematical Modeling Of Flashover
- 8. Staying Engaged with Experimental Study And Mathematical Modeling Of Flashover
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Experimental Study And Mathematical Modeling Of Flashover
- 9. Balancing eBooks and Physical Books Experimental Study And Mathematical Modeling Of Flashover
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Experimental Study And Mathematical Modeling Of Flashover
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Experimental Study And Mathematical Modeling Of Flashover
 - $\circ\,$ Setting Reading Goals Experimental Study And Mathematical Modeling Of Flashover
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Experimental Study And Mathematical Modeling Of Flashover
 - Fact-Checking eBook Content of Experimental Study And Mathematical Modeling Of Flashover
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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