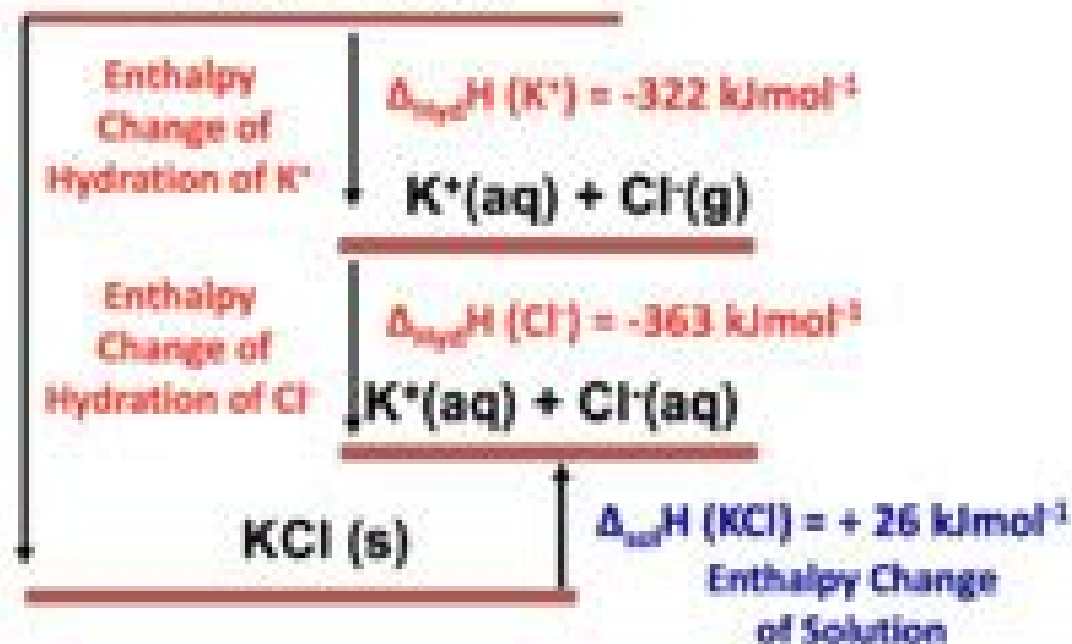


# Calculating Lattice Enthalpies from $\Delta_{\text{sol}}H$ & $\Delta_{\text{hyd}}H$

If we know  $\Delta_{\text{sol}}H$  &  $\Delta_{\text{hyd}}H$  we can calculate  $\Delta_{\text{LE}}H$

$K^+(g) + Cl(g)$

Lattice Enthalpy  
 $\Delta_{\text{LE}}H (KCl) = ?$



$$\begin{aligned}\Delta_{\text{hyd}}H (\text{metal}) + \Delta_{\text{hyd}}H (\text{non-metal}) &= \Delta_{\text{LE}}H + \Delta_{\text{sol}}H \\ \Delta_{\text{LE}}H &= \Delta_{\text{hyd}}H (\text{metal}) + \Delta_{\text{hyd}}H (\text{non-metal}) - \Delta_{\text{sol}}H \\ &= -322 - 363 - 26 \\ &= -711 \text{ kJmol}^{-1}\end{aligned}$$

# Enthalpy Of Solution Cacl

**Victor M. Corman**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

**Enthalpy Of Solution Cacl:**

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In some sort of consumed by monitors and the ceaseless chatter of fast communication, the melodic beauty and psychological symphony produced by the written term often fade into the backdrop, eclipsed by the persistent noise and disruptions that permeate our lives. Nevertheless, situated within the pages of **Enthalpy Of Solution Cacl** a marvelous literary treasure full of organic emotions, lies an immersive symphony waiting to be embraced. Crafted by a masterful musician of language, that interesting masterpiece conducts viewers on an emotional journey, skillfully unraveling the concealed tunes and profound affect resonating within each carefully crafted phrase. Within the depths of the emotional analysis, we shall explore the book is central harmonies, analyze their enthralling publishing design, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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