

# The Diary of Anne Frank TEST

1. The word 'inherent' means:  
a. bothersome, irritating  
b. quiet, restrained  
c. sneaky, secretive  
d. essential, basic, natural
2. The word 'obtrusive' means:  
a. to set free  
b. easily seen, noticeable  
c. sneaky, secretive  
d. essential, basic, natural
3. The word 'conspicuous' means:  
a. bothersome, irritating  
b. easily seen, noticeable  
c. hate, disgust, extreme dislike  
d. to make strong
4. When Anne and her mother disagree that is an example of which type of conflict?  
a. person vs. self  
b. person vs. person  
c. person vs. society  
d. person vs. nature
5. Jewish people being forced into hiding or sent to concentration camps is an example of which type of conflict?  
a. person vs. self  
b. person vs. person  
c. person vs. society  
d. person vs. nature
6. Which is NOT a precaution the families in the annex need to take?  
a. keeping curtains closed at all times  
b. only using W.C. at certain times  
c. not reading books or learning their school work  
d. being silent all day long
7. Which is true about Anne's relationship with Peter?  
a. They start out arguing and end up really liking each other  
b. They never learn to get along  
c. They are friends from the very start and always get along  
d. They begin getting along, but end up hating each other

# Diary Of Anne Frank Multiple Choice Questions

**Roman Wölfel**



## **Diary Of Anne Frank Multiple Choice Questions:**

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web solution 1 the two half reactions balanced as if in acidic solution  $2\text{CN}^- + \text{Au} \rightarrow \text{Au} + \text{CN}^- + 2\text{e}^-$   $2\text{H}_2\text{O}_2 \rightarrow 2\text{H}_2\text{O} + \text{O}_2 + 2\text{e}^-$  make electrons equal convert to basic solution  $4\text{CN}^- + 2\text{Au} \rightarrow 2\text{Au} + \text{CN}^- + 2\text{e}^-$  multiplied by a factor of 2  $2\text{e}^- + 2\text{H}_2\text{O}_2 \rightarrow \text{O}_2 + 2\text{OH}^-$  the final answer

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web we can balance it for charge by adding two electrons to the right side of the equation so that the net charge on each side is 0 oxidation ni s ni a 2 a q 2 e now that the oxidation half reaction is balanced it tells us that two electrons are produced for every atom of nickel oxidized but where do those electrons go

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