Determination of the Solubility-Product Constant for a Sparingly Soluble Salt

A. Preparation of a Calibration Curve Initial [CrO₄²⁻] 0.0014

Volume of 0.0024 M K ₂ CrO ₄		Total volume	[CrO ₄ ² -]	Absorbance
1	1 1	100 nl		0.122
2.	5	100 aL		0.625
Э.	10 06	100 1		1.788
4.3	15 -L	100 AL	1	1.955

Molar absorption coefficient for [CrO₄²⁻]

Error for molar absorption coefficient

B. Determination of the Solubility-Product Constant

	Absorbance	[CrO4 ²⁻]	[Ag*]	K_{sp} of $Ag_2CrO_4 = [CrO_4^2][Ag^+]$
1.	1.33 /		_	
3.	1.542.			

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