solvent

increases

solute

Overview Solutions

Directions: Use the words listed below to complete the paragraphs. Some words will be used more than once.

decreases	temperature	nonpolar	polar
electrolyte	nonelectrolyte	higher	lower
In a solution, the 1		does the dissolving and the	
2	is the substance	being dissolved. For	solids dissolving in
liquids, the speed of dissolving 3.		with increased surface	
area, with increased	temperature, and with	stirring. For gases in	liquids, the speed o
solubility 4.	bility 4 with increased temperature but 5.		
	with increased pr	essure.	
In general, polar	solutes dissolve in pol-	ar solvents and do no	t dissolve in
6	solvents. Usually, nonpolar solvents dissolve nonpolar		
solutes and do not d	issolve 7.	solutes.	
Solubility is usu	ally expressed as the ma	aximum number of g	rams of
8	that will dissolve in 100 grams of solvent at a certain		
9			
A substance that	forms charged ions in	solution and can con	duct electricity is
called a(n) 10.	A(n) 11	does not
ionize in water and	cannot conduct electric	ity. The boiling point	of a water solution
of any substance is 1:	than the boiling point of water.		
The freezing point of	a water solution of any	substance is 13	
than the freezing poi	nt of water.		

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