



Determination Of % Available Chlorine In Bleaching Powder

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To determine the available chlorine in the given sample of bleaching powder by the iodometric method.

Introduction

- · Bleaching powder, also known as chlorinated lime
- yellowish-white powder having a smell of chlorine and is readily soluble in water
- It is prepared by passing chlorine gas over slaked lime at a temperature of 35-45° and consists of a mixture
 of calcium hypochlorite Ca(OCl₂) and calcium chloride CaCl₃, in addition some amount of free slaked lime
 i.e. Ca(OH)₂ H₂O is generally present
- Ca(OCl₂) is responsible for the bleaching action of bleaching powder. On treatment with glacial acetic acid, it liberates chlorine gas (Cl₂) as per the following reaction.

$$Ca(OCI,)+2CH,COOH \rightarrow Ca(CH,COO), +H,O+CI,$$

 The amount of chlorine liberated by the action of an acid on bleaching powder (CaOCl₂) is termed as available chlorine. The chlorine content of bleaching powder varies from 35 - 40%.





- bleaching powder also have strong germicidal and disinfectant properties.
- It finds application as a disinfectant for drinking water or swimming pool water.

Determination Of Available Chlorine In Bleaching Solution

Arturo Cuomo

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information on theoretical and practical areas

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