

# CLR via C#

## Fourth Edition

 Developer Reference



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web the first two equations have been balanced for you alpha particle  $\alpha$  2 4 beta particle  $\beta$  gamma ray  $\gamma$  proton p neutron n 0 positron  $\beta^+$  0 1 30 1 65 1 2 63 2 15 0 30 0 65 30 9 29 2 2 12 6 15 32 30

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web we can write a balanced nuclear equation for this reaction as let s do a practice example identify the product x by writing a balanced nuclear equation  $^{137}_{55}\text{Cs} \rightarrow ^{137}_{56}\text{Ba} + x$  because the mass number is the same on both sides of the equation x must have a

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web identify the missing particle in the following nuclear reaction  $^{1}_{0}\text{n} + ^{235}_{92}\text{U} \rightarrow ^{137}_{52}\text{Te} + ^{96}_{40}\text{Zr} + ^{1}_{0}\text{n}$

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web answer 4 correct in the reactants and products must be the same part b what is the value of in the following nuclear reaction 6 5i 9 express your answer as an integer hint 1 how to approach the problem use rule 2 the total of the subscripts atomic numbers mathematically this can be written as answer

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