

# SALT ANALYSIS

1) For  $MgSO_4$  - magnesium Sulphate:-

**Aim:-** To analyze the given salt for an anion and one cation present in it.

**Material Required:-** Boiling Tubes, Test Tubes, Test Tube Holder, Test Tube Stand, delivery tube, Corks, Filter paper, X-ray paper.

**Procedure:-**

Sr.	Test	Observation	Inference
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i)	Colour	White	
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ii)	Odour	No odour	
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iii)	Flame test	Colourless	
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iv)	Test with $H_2SO_4$ (Salt + conc. $H_2SO_4$ )	No gas evolved	
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v)	Test with $H_2SO_4$ (Salt + conc. $H_2SO_4$ )	No gas evolved	
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ii) For  $Fe^{3+}Cl^-$ , Ferric chloride :-

**Aim:-** To analyze the given salt for an anion and one cation present in it.

**Material Required:-** Boiling Tubes, Test Tubes, Test Tube Holder, Test Tube Stand, delivery tube, Corks, Filter paper, X-ray paper.

**Procedure:-**

Sr.	Test	Observation	Inference
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		Brown reddish	$Fe^{3+}$ may be present
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		Colourless	$NO_3^-$ , $S^{2-}$ , $CO_3^{2-}$ are absent
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			$Ca^{2+}$ , $S^{2-}$ , $SO_4^{2-}$ , $NO_3^-$ are absent $Cl^-$ may be present
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Class 12th

Chemistry Practical File

Experiment - SALT ANALYSIS

5 most Important Salt

Handwritten PDF Notes

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