

1-6, 8, 9, 11

## Hydrolysis of Salts and pH of Buffer Solutions

## Pre-lab Questions

Before beginning this experiment in the laboratory, you should be able to answer the following questions.

1. Define Brønsted-Lowry acids and bases.

According to Brønsted-Lowry definitions, an acid is a proton donor and a base is a proton acceptor.

2. Which of the following ions will react with water in a hydrolysis reaction:  $K^+$ ,  $Ba^{2+}$ ,  $Cu^{2+}$ ,  $Zn^{2+}$ ,  $F^-$ ,  $SO_3^{2-}$ , and  $Cl^-$ ?

$Cu^{2+}$ ,  $Zn^{2+}$ ,  $F^-$ , and  $SO_3^{2-}$  will undergo hydrolysis.

3. For those ions in question 2 that undergo hydrolysis, write net ionic equations for the hydrolysis reaction.

Example:  $Cu^{2+}(aq) + 2H_2O(l) \rightleftharpoons Cu(OH)_2(s) + 2H^+(aq)$ .

4. The  $K_a$  for HCN is  $4.9 \times 10^{-10}$ . What is the value of  $K_b$  for  $CN^-$ ?

$$K_b = \frac{1.0 \times 10^{-14}}{4.9 \times 10^{-10}} = 2.0 \times 10^{-5}$$

5. What are the conjugate base and conjugate acid of  $H_2PO_4^-$ ?

The conjugate base is  $HPO_4^{2-}$  and the conjugate acid is  $H_3PO_4$ .

6. From what acid and what base were the following salts made:  $K_2SO_4$ ,  $NH_4Cl$ , and  $Ca(NO_3)_2$ ?

For  $Ca(NO_3)_2$ , the salt is made from  $Ca(OH)_2$  and  $HNO_3$ .

7. Define the term *salt*.

# Ph Properties Of Buffer Solutions Answer Key Pre Lab

**Colin Poole**



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**Green Chemistry Laboratory Manual for General Chemistry** Sally A. Henrie, 2015-03-18 Green chemistry involves designing novel ways to create and synthesize products and implement processes that will eliminate or greatly reduce negative environmental impacts Providing educational laboratory materials that challenge students with the customary topics found in a general chemistry laboratory manual this lab manual enables students to see how green chemistry principles can be applied to real world issues Following a consistent format each lab experiment includes objectives prelab questions and detailed step by step procedures for performing the experiments Additional questions encourage further research about how green chemistry principles compare with traditional more hazardous experimental methods **40 Inquiry Exercises for**

**the College Biology Lab** A. Daniel Johnson, 2009 Drawing from the author's own work as a lab developer coordinator and instructor this one of a kind text for college biology teachers uses the inquiry method in presenting 40 different lab exercises that make complicated biology subjects accessible to major and nonmajors alike The volume offers a review of various aspects of inquiry including teaching techniques and covers 16 biology topics including DNA isolation and analysis properties of enzymes and metabolism and oxygen consumption Student and teacher pages are provided for each of the 16 topics

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Hoppe-Seyler, Konrad Lang, Günther Siebert, Hans Thierfelder, 2019-06-12 **Chemical Principles in the Laboratory** Lester R. Morss, Robert S. Boikess, 1978 Nuclear Science Abstracts, 1973 NSA is a comprehensive collection of international nuclear science and technology literature for the period 1948 through 1976 pre dating the prestigious INIS

database which began in 1970 NSA existed as a printed product Volumes 1 33 initially created by DOE s predecessor the U S Atomic Energy Commission AEC NSA includes citations to scientific and technical reports from the AEC the U S Energy Research and Development Administration and its contractors plus other agencies and international organizations universities and industrial and research organizations References to books conference proceedings papers patents dissertations engineering drawings and journal articles from worldwide sources are also included Abstracts and full text are provided if available

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O Phillips, Peter A Williams, 2009-10-21 The latest volume in the successful Special Publication Series captures the most recent research findings in the field of food hydrocolloids The impressive list of contributions from international experts includes topics such as Hydrocolloids as dietary fibre The role of hydrocolloids in controlling the microstructure of foods The characterisation of hydrocolloids Rheological properties The influence of hydrocolloids on emulsion stability Low moisture systems Applications of hydrocolloids in food products Gums and Stabilisers for the Food Industry 12 with its wide breadth of coverage will be of great value to all who research produce process or use hydrocolloids both in industry and academia

**Industrial Laboratory**, 1958      **Energy Research Abstracts**, 1979      *EPA Publications Bibliography* United States.  
Environmental Protection Agency, 1985      *Selected Water Resources Abstracts*, 1990

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