

1-6, 8, 9, 11

Hydrolysis of Salts and pH of Buffer Solutions

Pre-lab Questions

Before beginning this experiment in the laboratory, you should be able to answer the following questions.

1. Define Brønsted-Lowry acids and bases.

According to Brønsted-Lowry definitions, an acid is a proton donor and a base is a proton acceptor.

2

- Which of the following ions will react with water in a hydrolysis reaction:

K^+ , Ba^{2+} , Cu^{2+} , Zn^{2+} , F^- , SO_3^{2-} , and Cl^- ?

Cu^{2+} , Zn^{2+} , and SO_3^{2-} will undergo hydrolysis.

3

- For those ions in question 2 that undergo hydrolysis, write net ionic equations for the hydrolysis reaction.

Example: $Cu^{2+}(aq) + 2H_2O(l) \rightleftharpoons Cu(OH)_2(s) + 2H^+(aq)$.

4

- The K_a for HCN is 4.9×10^{-10} . What is the value of K_b for CN^- ?

$$K_b = \frac{1.0 \times 10^{-14}}{4.9 \times 10^{-10}} = 2.0 \times 10^{-5}$$

5. What are the conjugate base and conjugate acid of $H_2PO_4^-$?

The conjugate base is HPO_4^{2-} and the conjugate acid is H_3PO_4 .

6

- From what acid and what base were the following salts made: K_2SO_4 , NH_4Cl , and $Ca(NO_3)_2$?

For K_2SO_4 , $CaSO_4$ may be made from $Ca(OH)_2$ and H_2SO_4 .

7

- Define the term *salt*.

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Colin Poole



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Green Chemistry Laboratory Manual for General Chemistry Sally A. Henrie, 2015-03-18 Green chemistry involves designing novel ways to create and synthesize products and implement processes that will eliminate or greatly reduce negative environmental impacts Providing educational laboratory materials that challenge students with the customary topics found in a general chemistry laboratory manual this lab manual enables students to see how green chemistry principles can be applied to real world issues Following a consistent format each lab experiment includes objectives prelab questions and detailed step by step procedures for performing the experiments Additional questions encourage further research about how green chemistry principles compare with traditional more hazardous experimental methods **40 Inquiry Exercises for**

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multiscale approach presenting both micro semi micro and macroscale techniques The manual is easy to navigate with all relevant techniques found as they are needed Cutting edge subjects such as HPLC bioorganic chemistry multistep synthesis and more are presented in a clear and engaging fashion **Methods in Biotechnology** Seung-Beom Hong, M. Bazlur

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Hoppe-Seyler, Konrad Lang, Günther Siebert, Hans Thierfelder, 2019-06-12 **Chemical Principles in the Laboratory** Lester R. Morss, Robert S. Boikess, 1978 Nuclear Science Abstracts, 1973 NSA is a comprehensive collection of

international nuclear science and technology literature for the period 1948 through 1976 pre dating the prestigious INIS

database which began in 1970 NSA existed as a printed product Volumes 1 33 initially created by DOE s predecessor the U S Atomic Energy Commission AEC NSA includes citations to scientific and technical reports from the AEC the U S Energy Research and Development Administration and its contractors plus other agencies and international organizations universities and industrial and research organizations References to books conference proceedings papers patents dissertations engineering drawings and journal articles from worldwide sources are also included Abstracts and full text are provided if available

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Industrial Laboratory, 1958 **Energy Research Abstracts**, 1979 *EPA Publications Bibliography* United States.
Environmental Protection Agency, 1985 *Selected Water Resources Abstracts*, 1990

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