Chemistry Periodic Table Test Chapter 6 Chem IH

Matching Match each item with the correct statement below.

| | | electronegativity ionization energy electrons metal | c. f. g. h. | transition metal periodic law group period |
|-------|---------|---|----------------------|---|
| | 1. | vertical column in the periodic table | | |
| | 2. | | | |
| | 3. | 사용하게 되었습니다. 그리고 있는 이 사용이 되었습니다. 그리고 있는데 하는데 이 사용이 되었습니다. 이 사용이 되었습니다. 그리고 있는데 사용이 사용이 되었습니다. | | |
| | 4. | 프루트 : 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 | | |
| | 5. | energy required to remove an electron from an atom | | |
| Month | tiple C | hoice | | |
| | | letter of the choice that best completes th | e statemen | at or answers the question. |
| | 6. | . What is another name for the main group elements? | | |
| | | a. s block elements | C. | s & p block elements |
| | | b. p block elements elements | d. | d block elements |
| | 7. | What is another name for the transition r | metals? | |
| | | a. noble gases | e. | p block elements elements |
| | | b. s block elements elements | d. | d block elements |
| | 8. | 6. Who arranged the elements according to atomic mass and used the arrangement to predict the properties missing elements? | | |
| | | a. Henry Moseley | 421. | John Dalton |
| | | b. Antoine Lavoisier | d. | Dmitri Mendeleev |
| | 9. | Which of the following categories include | les the maj | ority of the elements? |
| | | a. metalloids | | metals |
| | | b, liquids | d. | nonmetals |
| | 10. | electrons given correctly? | bol of the | element, the number of protons, and the number of |
| | | a. In, 49 protons, 49 electrons | 42. | Cs, 55 protons, 132.9 electrons |
| | | b. Zn, 30 protons, 60 electrons | d. | F, 19 protons, 19 electrons |
| | 11. | The atomic number of an element is the | total namb | er of which particles in the nucleus? |
| | | a. neutrons | e. | electrons |
| | | b. protons | | protons and electrons |
| | 12. | What element has the electron configura | tion $1s^22s^2$ | $^{1}2p^{6}3s^{2}3p^{2}$? |
| | 1000000 | a. nitrogen | e. | silicon |
| | | b. selenium | d. | silver |
| | 13. | Which subatomic particle plays the great | est part in | determining the properties of an element? |
| | | a. proton | | neutron |
| | | b. electron | d. | none of the above |
| | 114. | Which of the following elements is a train | ssition met | tal? |
| | | a. cesium | e. | tellurium |
| | | b. copper | d. | tin |

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KJ Lindholm-Leary

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