

Name:

Kyleen Mitchell

Date:

11-5-20

#### Student Exploration: Balancing Chemical Equations

Directions: Follow the instructions to go through the simulation, Respond to the questions and prompts in the orange boxes.

Vocabulary: coefficient, combustion, compound, decomposition, double replacement, element, molecule, product, reactant, single replacement, subscript, synthesis

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

The scouts are making s'mores out of toasted marshmallows, chocolate, and graham crackers.

1. What is wrong with the image below?

What is wrong with the image below is that there are too many marshmallows and not enough chocolate and graham crackers.



Assuming a s'more requires two graham crackers, one marshmallow, and one piece of chocolate, how many s'mores could you make with the ingredients shown?

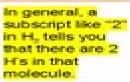
Assuming a s'more requires two graham crackers, one marshmallow, and one piece of chocolate, you could make 2 s'mores with the ingredients shown.

#### Gizmo Warm-up

In a chemical reaction, reactants interact to form products. This process is summarized by a chemical equation. In the Balancing Chemical Equations Gizmo, look at the floating molecules below the initial reaction:  $H_2 + O_2 \square H_2O$ .

- How many atoms are in a hydrogen molecule (H<sub>2</sub>)?
- How many atoms are in an oxygen molecule (O<sub>2</sub>)?
- How many hydrogen and oxygen atoms are in a water molecule (H<sub>2</sub>O)?
- In general, what does a subscript (such as the "2" in H<sub>2</sub>) tell you about the molecule?







# **Balancing Chemical Equations Gizmo Answers**

**James Burke** 

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