

**Chemistry – Unit 4: Motion****Ballon's Playground**

In this case, 100% oxygen, through the use of a chemical reaction, was used to produce some of the components of the original balloon, including the nitrogen which allows it to expand. In production of this mixture:

https://www.youtube.com/watch?v=9486t8u3u00 will allow you to experience the reaction that was used to produce the nitrogen gas. Answer the questions on this worksheet and keep track of your progress as this video is played.

**Part 1 – Planning**

| Task                         | Start | Finish | Days Left |
|------------------------------|-------|--------|-----------|
| Read of each student's notes | 11/11 | 11/11  | 1 day     |
| Worksheet part 1             | 11/11 | 11/11  | 1 day     |

- What happened to the mass of the material in the flask as it was heated?  
Is it always the same?
- What did you note about the volume of the gas produced and the volume of the flask before it was heated?
- Does it depend on the temperature?
- Does the relationship between the volume of gas produced and the mass of the metal that was heated?
- What are always the same in each flask? How much of the substance that is used as gas?

**Part 2 – Calculating**

Use the information from the table and find values for oxygen and nitrogen in each of the trials using the following formulae:

|                      | Mass 1 | Mass 2 | Mass 3 |
|----------------------|--------|--------|--------|
| Mass of gas produced | 1      | 1.50   | 2.25   |
| Mass of flask before | 10     | 10     | 10     |
| Mass of flask        | 11     | 11.5   | 12.25  |
| Mass of flask + gas  | 11     | 11.5   | 12.25  |
| Mass of nitrogen gas | 1      | 1      | 1      |
| Mass of oxygen gas   | 1      | 1      | 1      |

# Modeling Chemistry Dalton Playhouse Notes Answers

**James Silk Buckingham, John  
Sterling, Frederick Denison  
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Hervey, William Hepworth  
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Rendall, John Middleton Murry**

## **Modeling Chemistry Dalton Playhouse Notes Answers:**

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