# Inkjet printing of nanosized silver colloids

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#### Abstract

A water-based conducting ink, composed of well dispersed nano-silver particles, has been successfully inkjet printed using an ordinary commercial printer. The silver colloids with diameter around 50 nm were dispersed in a water and diethylene glycol cosolvent system. The 25 wt% silver ink had a viscosity of about 7.4 cP and surface tension of 33.5 dyn cm<sup>-1</sup> at 20 °C, which is appropriate for printing jobs. Continuous and smooth lines of 130 μm width could be printed on ordinary glass substrates using an Epson R210 printer. After baking at 260 °C for 3 min, these lines exhibited a resistivity of 1.6 × 10<sup>-5</sup> Ω cm, which could serve as conducting lines for electronic applications.

(Some figures in this article are in colour only in the electronic version)

#### 1. Introduction

In the electronic industry, fabrication of conductive lines (circuit) is vital and inevitable. Traditionally, for example, electroplating and etching processes accompanied by lithography technology are widely adopted in the printing circuit board (PCB) industry for manufacturing its circuits. However, this method is not only time consuming but also very complicated, because many steps are required to construct one layer of circuit. Moreover, the conventional electroplating and etching processes also produce large quantities of waste water. For these reasons, the development of convenient and fast processing techniques to fabricate conductive lines has attracted more and more attention in recent years.

One of the most promising alternatives is the ink-jet printing technology, by which the conductive lines can be drawn (i.e. printed) onto the substrate in one step. There are two important components of the ink-jet printing technology: one is the mechanical system, i.e. the printer, and the other is the conductive material, i.e. the ink. Studies focused on the former item discussed the formation of droplets, accurate control of dropping location, jetting parameters, accurate control of dropping location, jetting parameters, are [1–5]. On the research of conductive ink, several materials were studied including molten metal, conductive polymer, and metallic nanoparticle suspension [6]. Among these, metallic nanoparticle suspension gained significant interest in recent years because it can be operated at room

temperature [7] as compared to the molten metal, and also has a better performance in terms of conductivity than conductive polymers on the other hand.

Viscosity and surface tension are the two most important properties of general inks [8]. Size, dispersion and stability of metallic nanopurticles are also crucial for the conductive ink system, because the nozzle of the printer would be clogged with larger particles or due to the aggregation of nanoparticles. In addition, the physical properties of metallic nanoparticles undoubtedly affect the performance of the product to a great extent. Due to their high conductivity and thermal stability, gold and silver nanoparticle suspensions were widely used in the studies of conductive ink. Some operating parameters in these studies are listed in table 1 19-141 for comparison. In the work by Fuller et al in 2002 [9]. Ag nanoparticles suspended in ar-terpineol were printed by the piezo-driven ink-jet printer onto a heated substrate at 100-300 °C. After sintering at 300 °C for 10 min, the fabricated lines showed a resistivity of approximately 3  $\mu\Omega$  cm, about twice that of bulk silver (1.51 p: \$2 cm). However, in this case a-terpineol was used as the solvent and it may cause the problem of volatile organic compounds (VOCs) in realistic industrial applications. In the work of Bieri et al in 2004 [12], Au nanoparticles suspended in toluene were used. Conductive ink deoplets were generated by the piezo-driven ink-jet printer and deposited onto a moving substrate to form continuous lines. Then focused laser irradiation was utilized to evaporate the solvent and sinter these Au nanoparticles. The result showed that micro-lines

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fish are not abundant and a beginner may come across as many as an older fisher but we speak of lochs where there are fish to be caught and where each has a fair chance Again it is said that the boatman has as much to do with catching trout in a loch as the angler Well we dont deny that In an untried loch it is necessary to have the guidance of a good boatman but the same argument holds good as to stream fishing The Theory and Application of Colloidal Behavior Robert Herman Boque, 1924 Silver Colloids Ronald J. Gibbs, 1999 The Colloidal Salts Harry Boyer Weiser, 1928 Und Die Photohaloide Hinricus Lüppo-Cramer, Mathew Carey Lea, 2023-07-18 Lea besch ftigt sich in diesem Buch mit der Herstellung von kolloidalem Silber und dessen Verwendung als Foto Aufnahmematerial Er beschreibt die chemischen Prozesse und gibt dem Leser wertvolle Hinweise zur praktischen Anwendung Ein wichtiges Werk fr alle Fotografen und Chemiker This work has been selected by scholars as being culturally important and is part of the knowledge base of civilization as we know it This work is in the public domain in the United States of America and possibly other nations Within the United States you may freely copy and distribute this work as no entity individual or corporate has a copyright on the body of the work Scholars believe and we concur that this work is important enough to be preserved reproduced and made generally available to the public We appreciate your support of the preservation process and thank you for being an important part of keeping this knowledge alive and relevant Kolloides Silber Und Die Photohaloide Hinricus Lüppo-Cramer, Mathew Carey Lea, 2023-07-18 Lea besch ftigt sich in diesem Buch mit der Herstellung von kolloidalem Silber und dessen Verwendung als Foto Aufnahmematerial Er beschreibt die chemischen Prozesse und gibt dem Leser wertvolle Hinweise zur praktischen Anwendung Ein wichtiges Werk fralle Fotografen und Chemiker This work has been selected by scholars as being culturally important and is part of the knowledge base of civilization as we know it This work is in the public domain in the United States of America and possibly other nations Within the United States you may freely copy and distribute this work as no entity individual or corporate has a copyright on the body of the work Scholars believe and we concur that this work is important enough to be preserved reproduced and made generally available to the public We appreciate your support of the preservation process and thank you for being an important part of keeping this knowledge alive and relevant Therapeutical Colloids ,1913 **Photogeneration and Degradation of Colloidal Silver Particles** Colloidal Dispersions Earl K. Fischer, 1950 Recent Clinical Literature on Collargolum, Kathryn Elder Foti, 2006 Soluble Metallic (colloidal) Silver-heyden Schering & Glatz Firm, 1913

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- 1. Understanding the eBook Download Pdf Colloidal Silver Silver Colloids
  - The Rise of Digital Reading Download Pdf Colloidal Silver Silver Colloids
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Download Pdf Colloidal Silver Silver Colloids
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Download Pdf Colloidal Silver Silver Colloids
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Download Pdf Colloidal Silver Silver Colloids
  - Personalized Recommendations
  - Download Pdf Colloidal Silver Silver Colloids User Reviews and Ratings

- Download Pdf Colloidal Silver Silver Colloids and Bestseller Lists
- 5. Accessing Download Pdf Colloidal Silver Silver Colloids Free and Paid eBooks
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  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Download Pdf Colloidal Silver Silver Colloids
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  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Download Pdf Colloidal Silver Silver Colloids
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  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Download Pdf Colloidal Silver Silver Colloids
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  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Download Pdf Colloidal Silver Silver Colloids
  - Fact-Checking eBook Content of Download Pdf Colloidal Silver Silver Colloids
  - Distinguishing Credible Sources

- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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