

Top 10 (Basic) Mechanical Engineering Interview Questions & Answers

1) What is the second law of thermodynamics?

The second law of thermodynamics portrays that the complete entropy of a detached framework can never lessen over the long haul.

Q) What is the first law?

Energy is an absolute conserved value.

Q) What does the first law of thermodynamics state?

Heat added to a system (dq) is equal to the change in internal energy (du).

Q) What does the second law of thermodynamics state?

The total entropy of a system and its surroundings can never decrease, and the change in entropy is always greater than or equal to zero.

Q) What is the third law of thermodynamics?

The entropy of a perfect crystal is zero at absolute zero.

Entropy is a state function.

Entropy is a measure of disorder.

Q) What is the Carnot cycle?

It is a thermodynamic cycle that operates between two heat reservoirs at different temperatures.

Q) What is the efficiency of a Carnot cycle?

The efficiency of a Carnot cycle is the ratio of the work done by the cycle to the heat added to the cycle. It is the maximum efficiency possible for a heat engine operating between two heat reservoirs.

Q) What is the Clausius inequality?

The change in entropy is greater than or equal to the heat added divided by the temperature.

Q) What is the significance of the Carnot cycle?

It is a theoretical cycle that sets the upper limit for the efficiency of any heat engine.

Q) What is the significance of the second law of thermodynamics?

The second law of thermodynamics is a statement of the fact that the total entropy of a system and its surroundings can never decrease. It is a fundamental principle of physics that governs the behavior of all systems.

Basic Mechanical Engineering Interview Questions And Answers

Ian Pickup



Basic Mechanical Engineering Interview Questions And Answers:

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