

Balancing Chemical Equations

Once we have written a chemical equation, we must make sure to balance it! Balancing a chemical equation means to make sure that there are the same number of each type of atom on both sides of the chemical equation. So, if there are five carbons on the left side (reactants), there needs to be five carbons on the right side (products).

So, how do we balance an equation? Follow the following steps...

- Change **only** stoichiometric coefficients – those big numbers in front of the compounds (not subscripts in the formulas – that would change the identity of the compound!)
- Start with most complex
- If found in more than one place on each side, do it last
- Watch out for odd numbers (make them into even ones if needed)
- Check your work!

Now, try to balance the following equations. In addition, name the bolded compounds for some nomenclature practice!

- $\underline{\quad} \text{NH}_4\text{NO}_3(\text{s}) \rightarrow \underline{\quad} \text{N}_2\text{O}(\text{g}) + \underline{2} \text{H}_2\text{O}(\text{g})$
ammonium nitrate & dinitrogen monoxide
- $\underline{\quad} \text{CH}_3\text{CO}_2\text{H}(\text{aq}) + \underline{2} \text{O}_2(\text{g}) \rightarrow \underline{2} \text{CO}_2(\text{g}) + \underline{2} \text{H}_2\text{O}(\text{l})$
oxygen & carbon dioxide
- $\underline{2} \text{Pb}(\text{NO}_3)_2(\text{s}) \rightarrow \underline{2} \text{N}_2\text{O}_4(\text{g}) + \underline{2} \text{PbO}(\text{s}) + \underline{\quad} \text{O}_2(\text{g})$
lead(II) nitrate & dinitrogen tetroxide & lead(II) oxide
- $\underline{\quad} \text{SiO}_2(\text{s}) + \underline{4} \text{HF} \rightarrow \underline{\quad} \text{SiF}_4(\text{g}) + \underline{2} \text{H}_2\text{O}(\text{l})$
silicon dioxide & silicon tetrafluoride
- $\underline{8} \text{H}_2\text{S}(\text{aq}) + \underline{8} \text{Cl}_2(\text{g}) \rightarrow \underline{\quad} \text{S}_8(\text{s}) + \underline{16} \text{HCl}(\text{aq})$
chlorine
- $\underline{\quad} \text{H}_2\text{CO}_3(\text{aq}) + \underline{\quad} \text{Ba}(\text{OH})_2 \rightarrow \underline{\quad} \text{BaCO}_3(\text{s}) + \underline{2} \text{H}_2\text{O}(\text{l})$
barium hydroxide & barium carbonate
- $\underline{\quad} \text{P}_4(\text{s}) + \underline{6} \text{Br}_2(\text{l}) \rightarrow \underline{4} \text{PBr}_3(\text{l})$
phosphorous & phosphorous tribromide
- $\underline{\quad} \text{S}_8(\text{l}) + \underline{4} \text{Cl}_2(\text{g}) \rightarrow \underline{4} \text{S}_2\text{Cl}_2(\text{g})$
disulfur dichloride
- $\underline{2} \text{B}_5\text{H}_9(\text{l}) + \underline{12} \text{O}_2(\text{g}) \rightarrow \underline{5} \text{B}_2\text{O}_3(\text{s}) + \underline{9} \text{H}_2\text{O}(\text{g})$
pentaboron nonahydride & diboron trioxide
- $\underline{3} \text{H}_2\text{SO}_4(\text{aq}) + \underline{2} \text{Al}(\text{OH})_3(\text{s}) \rightarrow \underline{\quad} \text{Al}_2(\text{SO}_4)_3(\text{aq}) + \underline{6} \text{H}_2\text{O}(\text{l})$
aluminum hydroxide & aluminum sulfate
- $\underline{2} \text{Al}(\text{s}) + \underline{3} \text{Br}_2(\text{l}) \rightarrow \underline{2} \text{AlBr}_3(\text{s})$
aluminum bromide
- $\underline{\quad} (\text{NH}_4)_2\text{Cr}_2\text{O}_7(\text{s}) \rightarrow \underline{\quad} \text{N}_2(\text{g}) + \underline{\quad} \text{Cr}_2\text{O}_3(\text{s}) + \underline{4} \text{H}_2\text{O}(\text{l})$
ammonium dichromate & chromium(III) oxide
- $\underline{\quad} \text{S}_8(\text{s}) + \underline{24} \text{F}_2(\text{g}) \rightarrow \underline{8} \text{SF}_6(\text{g})$
sulfur hexafluoride
- $\underline{\quad} \text{Zn}(\text{s}) + \underline{\quad} \text{Pb}(\text{NO}_3)_2(\text{aq}) \rightarrow \underline{\quad} \text{Zn}(\text{NO}_3)_2(\text{aq}) + \underline{\quad} \text{Pb}(\text{s})$
zinc nitrate

Balancing Chemical Equations Multiple Choice Test

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Passing Oxbridge Admissions Tests Rosalie Hutton, Glenn Hutton, 2013-01-31 If you are applying to Oxford or Cambridge Universities you may have to take one of the Oxbridge specific admission tests This text provides the essential information you need to understand the format and structure of the tests along with vital practice in the sort of questions you will face The book covers in detail the Thinking Skills Assessment TSA for both institutions focusing on critical thinking and problem solving skills It includes a practice test with answers and explanations and also guidance on the Writing Task undertaken by applicants to Oxford The STEP and other subject specific tests are also examined Chapter-wise Topical Objective Study Package for CBSE 2022 Class 10 Term I Science Disha Experts, 2021-09-01 , **Goyal's ICSE Chemistry Question Bank with Model Test Papers For Class 10 Semester 2 Examination 2022** Goyal Brothers Prakashan, 2021-12-12 CISCE's Modified Assessment Plan for Academic Year 2021-22 Reduced and Bifurcated Syllabus for Semester 2 Examination Chapterwise Summary and Important Points Chapterwise Question Bank having all varieties of expected Questions with answers for Semester 2 Examination to be held in March April 2022 Specimen Question Paper Solved for Semester 2 Examination issued by CISCE 5 Model Test Papers based on the latest specimen question paper issued by CISCE for Semester 2 Examination to be held in March April 2022 Goyal Brothers Prakashan Olympiad Champs Science Class 10 with 5 Mock Online Olympiad Tests Disha Experts, 2017-07-04 OLYMPIADS Champs Class 10 Science is an attempt to guide and prepare students for Olympiad examinations The book will not only prepare the students for these examinations but will also help in developing a good aptitude and problem solving skills The book covers the complete science portion which is divided into three sections Physics Chemistry and Biology The book provides for each chapter important concepts followed by Multiple Choice Questions Exercises Concepts are summarized in the form of concept map at the end of each chapter Each chapter provides 2 levels of Exercises based on the level of difficulty Each exercise contains Simple MCQs Matching based MCQs statement based MCQs assertion reason based MCQs passage based MCQs and figure picture based MCQs The detailed solutions to the MCQs are provided at the end of each chapter 5 Online mock tests based on the different Olympiad exams are also provided along with the book This book will really prove to be an asset for Class 10 students as they hardly find any material which can help them in building a strong foundation Chemistry for the IB Diploma Exam Preparation Guide Steve Owen, Chris Martin, 2015-06-25 Chemistry for the IB Diploma Second edition covers in full the requirements of the IB syllabus for Chemistry for first examination in 2016 **CliffsNotes AP Chemistry 2021 Exam** Angela Woodward Spangenberg, 2020-09-29 CliffsNotes AP Chemistry 2021 Exam gives you exactly what you need to score a 5 on the exam concise chapter reviews on every AP Chemistry subject in depth laboratory investigations and full length model practice exams to prepare you for the May 2021 exam Revised to even better reflect the new AP Chemistry exam this test prep guide includes updated content tailored to the May 2021 exam Features of the guide focus on what AP Chemistry test takers need

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CliffsNotes AP Chemistry Bobrow Test Preparation Services, 2009-02-09. The book itself contains chapter-length subject reviews on every subject tested on the AP Chemistry exam, as well as both sample multiple-choice and free-response questions at each chapter's end. Two full-length practice tests with detailed answer explanations are included in the book.

A New Approach to ICSE Chemistry for Class X (A.Y. 2023-24) Onward V. K. Sally, 2023-05-20. A New Approach to ICSE Chemistry for Class X has been revised in accordance with the latest syllabus prescribed by the Council for the Indian School Certificate Examination, New Delhi, for Class X. The main strength of the book lies in the presentation of scientific

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