Date _____ Class

THE NATURE OF COVALENT BONDING

Section Review

Objectives

- State a rule that usually tells how many electrons are shared to form a covalent bond
- Describe how electron dot formulas are used
- Predict when two atoms are likely to be joined by a double or a triple covalent bond
- Distinguish between a single covalent bond and other covalent bonds
- Describe how the strength of a covalent bond is related to its bond dissociation energy
- Describe how resonance structures explain bonding
- Identify some exceptions to the octet rule

Vocabulary

- · single covalent bond
- structural formulas
- unshared pairs

- double covalent bonds
- triple covalent bonds.
- coordinate covalent bond
- polyatomic ion
- bond dissociation energy
- resonance structures

Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

When atoms share electrons to gain the ____1__ configuration of a noble gas, the bonds formed are 2 A 3 pair of valence electrons constitutes a ___4 __ covalent bond. Pairs of valence electrons that are not shared between atoms are called 5 Sometimes two or three pairs of electrons may be shared to give ___6__ covalent bonds. In some cases, only one of the atoms in a bond provides the pair of bonding electrons; this is a 7 8 is required to break covalent bonds between atoms. The total energy required to break the bond between two covalently bonded atoms is known as the ___9__.

When it is possible to write two or more valid electron dot formulas for a molecule or ion, each formula is referred to as a __

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2.	covalent	

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Covalent Bonding Section 1 Answers

Didier Musso

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Molecular Theory Of Water And Aqueous Solutions - Part 1: Understanding Water Arieh Ben-naim, 2009-07-17 The aim of this book is to explain the unusual properties of both pure liquid water and simple aqueous solutions in terms of the properties of single molecules and interactions among small numbers of water molecules It is mostly the result of the author's own research spanning over 40 years in the field of aqueous solutions. An understanding of the properties of liquid water is a prelude to the understanding of the role of water in biological systems and for the evolvement of life The book is targeted at anyone who is interested in the outstanding properties of water and its role in biological systems It is addressed to both students and researchers in chemistry physics and biology **Electronic Devices and Circuit Fundamentals,** Solution Manual Dale R Patrick, Stephen W. Fardo, Ray E. Richardson, Vigyan (Vigs) Chandra, 2023-05-26 Devices and Circuit Fundamentals is Chapter Outline Learning Objectives Key Terms Figure List Chapter Summary Formulas Answers to Examples Self Exams Glossary of Terms defined Solutions Manual to Accompany Inorganic Chemistry Alen Hadzovic, 2018 As you master each chapter in Inorganic Chemistry having detailed solutions handy allows you to confirm your answers and develop your ability to think through the problem solving process Essentials of Anatomy and Physiology Valerie C Scanlon, Tina Sanders, 2014-11-25 Leading the way for nearly 25 years with unsurpassed clarity content and completeness A student friendly writing style superb art program a wealth of learning opportunities in every chapter and online activities instill confidence every step of the way It's the perfect introduction to the world of anatomy NCERT Based Chapterwise Topicwise Solutions For 11th And 12th Class with Solved Papers (2005 -2023) with Notes for NEET-AIIMS Exam 2024 - Chemistry Dr K.G. Ojha, 2023-07-14 Excel in Chemistry for NEET AIIMS Exam 2024 with this comprehensive guide featuring objective NCERT based solutions solved papers and notes for classes 11th and 12th Objective NCERT From Prabhat Exam is an unparallel book designed on the complete syllabus of 11th and 12th NCERT textbook It is the leading choice of Toppers and the pinnacle for NEET exam along with NCERT This book is a must for NEET BOARDS CUET as it has questions extracted from each and every line of the NCERT textbook Extra Notes are added from experts to make it more understandable Chapter wise NCERT notes for quick yet thorough impactful revisions Tabular texts Illustrative diagrams in HD pages for understanding NCERT Based Topic wise MCQs from each of NCERT to get firm grip on concepts NCERT Exemplar Problem MCQs to develop a strong base go in depth Assertion Reason Case Based Questions HOTS to cover all question typologies Exam Archive including Previous years NEET other PMT exam's questions Practice Papers Model Test Papers to put final practice touch to your preparation 5 Mock Test to Make you an experienced player Answer keys hints and explanations are also added in the book for micro level understanding **Principles of Cell Biology** George Plopper, 2014-10-21 Written for undergraduate cell biology courses Principles of Cell Biology Second Edition provides students with the formula for understanding the fundamental concepts of cell biology This practical text focuses on the

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