

Identifying unknown using stoichiometry

Experiment 2

Experiment performed on 2/22/2021

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Lab partners: Antonio Alderete, Tiana Marie and
Allie Howard

Chemistry 1215L section 12

Hypothesis:

The only difference between sodium carbonate and sodium bicarbonate is the amount of CO₂ produced during the reaction. Since the CO₂ production is the only component of the reaction that would cause a significant change in pressure, the pressure measurements of the reactions of both unknown chemicals with HCl can provide a direct link to the stoichiometry of CO₂.

Chemicals and hazard:

Chemical name	Molecular formula	Molar mass	Amount used
Sodium chloride	NaCl	58.44 g/mL	'product'
Sodium carbonate	Na ₂ CO ₃	105.989 g/mL	25 mL
Hydrochloric acid	HCl	36.458 g/mL	50 mL
water	H ₂ O	18.01528 g/mL	'product'
Carbon dioxide	CO ₂	44.01 g/mL	'product'
Sodium hydrogen carbonate	NaHCO ₃	88.007 g/mL	25 mL

Identifying Unknown Solutions Chemistry Lab Report

Fitzgerald B. Bramwell



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