

Table 2. General tolerances and tolerances on dimensions with deviations entered against the dimension

Tolerance group from table 1	Code letter 1)	Nominal dimension range																							
		over up to	0 1	1 3	3 6	6 10	10 18	18 30	30 50	50 80	80 120	120 180	180 250	250 315	315 400	400 500	500 630	630 800	800 1000						
General tolerances																									
160	A		±0,28	±0,20	±0,20	±0,27	±0,42	±0,48	±0,67	±0,88	±0,78	±0,94	±1,15	±1,40	±1,80	±2,20	±2,75	±3,30	±4,10	±5,10	±6,30	±7,90	±10,00		
	B		±0,18	±0,20	±0,23	±0,27	±0,30	±0,36	±0,47	±0,58	±0,64	±0,84	±1,00	±1,20	±1,70	±2,10	±2,80	±3,30	±4,00	±5,00	±6,30	±7,90	±9,90		
150	A		±0,23	±0,25	±0,27	±0,30	±0,34	±0,38	±0,43	±0,49	±0,57	±0,88	±0,81	±0,87	±1,20	±1,30	±1,80	±2,20	±2,80	±3,40	±4,30	±5,30	±6,80		
	B		±0,13	±0,15	±0,17	±0,20	±0,24	±0,28	±0,33	±0,38	±0,47	±0,58	±0,71	±0,87	±1,10	±1,40	±1,70	±2,10	±2,70	±3,30	±4,30	±5,30	±6,80		
140	A		±0,20	±0,21	±0,22	±0,24	±0,27	±0,30	±0,34	±0,38	±0,43	±0,50	±0,60	±0,70	±0,85	±1,08	±1,28	±1,88	±1,80	±2,30	±2,80	±3,60	±4,80		
	B		±0,10	±0,11	±0,12	±0,14	±0,17	±0,20	±0,24	±0,28	±0,33	±0,40	±0,50	±0,60	±0,78	±0,98	±1,18	±1,48	±1,80	±2,30	±2,80	±3,60	±4,40		
130	A		±0,18	±0,19	±0,20	±0,21	±0,23	±0,25	±0,27	±0,30	±0,34	±0,38	±0,44	±0,51	±0,60	±0,70	±0,80	±1,10	±1,30	±1,80	±2,00	±2,50	±3,00		
	B		±0,08	±0,09	±0,10	±0,11	±0,12	±0,15	±0,17	±0,20	±0,24	±0,28	±0,34	±0,41	±0,50	±0,60	±0,80	±1,00	±1,20	±1,50	±1,80	±2,40	±2,90		
Tolerances on dimensions with deviations entered against the dimension																									
160	A		0,38	0,60	0,88	0,74	0,94	0,98	1,14	1,30	1,58	1,88	2,30	2,80	3,60	4,40	5,40	6,80	8,20	10,20	12,50	15,80	20,00		
	B		0,38	0,40	0,48	0,54	0,64	0,78	0,94	1,12	1,38	1,68	2,10	2,80	3,40	4,20	5,20	6,40	8,00	10,00	12,50	15,80	19,80		
150	A		0,48	0,80	1,24	0,60	0,88	0,78	0,98	0,98	1,14	1,38	1,62	1,94	2,40	3,00	3,80	4,40	5,60	6,80	8,80	10,80	13,20		
	B		0,28	0,30	0,34	0,40	0,48	0,58	0,68	0,78	0,94	1,18	1,42	1,74	2,20	2,80	3,40	4,20	5,40	6,80	8,40	10,40	12,80		
140	A		0,40	0,42	0,44	0,48	0,54	0,60	0,68	0,78	0,88	1,00	1,20	1,40	1,70	2,10	2,50	3,10	3,80	4,80	5,80	7,20	9,00		
	B		0,20	0,22	0,24	0,28	0,34	0,40	0,48	0,58	0,68	0,80	1,00	1,20	1,50	1,80	2,20	2,80	3,60	4,40	5,60	7,00	8,80		
130	A		0,36	0,38	0,40	0,42	0,48	0,50	0,54	0,60	0,68	0,78	0,88	1,02	1,20	1,50	1,80	2,20	2,80	3,20	3,80	4,80	6,00		
	B		0,16	0,18	0,20	0,22	0,28	0,30	0,34	0,40	0,48	0,58	0,68	0,82	1,00	1,30	1,60	2,00	2,40	3,00	3,70	4,70	5,80		
120	A		0,32	0,34	0,36	0,38	0,40	0,42	0,48	0,50	0,54	0,60	0,68	0,78	0,90	1,08	1,24	1,50	1,80	2,20	2,80	3,20	4,00		
	B		0,12	0,14	0,16	0,18	0,20	0,22	0,28	0,30	0,34	0,40	0,48	0,58	0,70	0,86	1,04	1,30	1,60	2,00	2,40	3,00	3,80		
110	A		0,18	0,20	0,22	0,24	0,28	0,28	0,30	0,32	0,38	0,40	0,44	0,50	0,58	0,68	0,88	1,18	1,40	1,70	2,10	2,80			
	B		0,08	0,10	0,12	0,14	0,18	0,18	0,20	0,22	0,28	0,30	0,34	0,40	0,48	0,58	0,70	0,86	1,08	1,30	1,60	2,00	2,80		
Precision engineering	A		0,10	0,12	0,14	0,16	0,20	0,22	0,24	0,28	0,28	0,31	0,38	0,40	0,50										
	B		0,05	0,06	0,07	0,08	0,10	0,12	0,14	0,18	0,18	0,21	0,25	0,30	0,40										

1) A for non-mould-related dimensions
B for mould-related dimensions

Din 16901 140

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web describe redox reactions in terms of electron transfer recall that oxidation involves loss of electrons and reduction involves gain of electrons represent simple redox reactions using half equations identify oxidising and reducing agents for simple redox reactions understand the reactivity of metals and metal displacement reaction

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web what is an oxidation reduction or redox reaction this is a reaction in which electrons are transferred between reactants what do the following terms mean oxidation oxidation is a loss of electrons reduction reduction is a gain of electrons what is a helpful way to remember this

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web write the oxidation and reduction reactions for each redox reaction the first one is done for you $23 \text{ Fe} + 2 \text{Co} \rightarrow \text{Fe}_2\text{O}_3 + \text{Co}_2\text{O}_3$
oxidation $\text{Co} \rightarrow \text{Co}_2\text{O}_3 + 2\text{e}^-$ reduction $\text{Fe}_2\text{O}_3 + 2\text{e}^- \rightarrow \text{Fe}$ $24 \text{ 3 Ag} + \text{Ni} \rightarrow \text{Ni}_3\text{O}_2 + 3 \text{ Ag}$ oxidation reduction $25 \text{ Cu} + 2 \text{ Pb} \rightarrow \text{Cu}_2\text{O} + \text{Pb}_2\text{O}_3$ oxidation reduction 26 O_2

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5co i2 5co2 i 2 o 5 5 c o i 2 5 c o 2

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