

8. Metallurgy question and answer

1. Write names.

a. Alloy of sodium with mercury.

Ans : Sodium amalgam, commonly denoted as Na(Hg), it is an alloy of mercury and sodium.

b. Molecular formula of the common ore of aluminium.

Ans : Bauxite($\text{Al}_2\text{O}_3 \cdot 2\text{H}_2\text{O}$) is the common ore of aluminium.

c. The oxide that forms salt and water by reacting with both acid and base.

Ans: Metal oxides which react with both acids as well as bases to produce salts and water are also known as amphoteric oxides. For example: Al_2O_3 is an amphoteric oxide. An amphoteric compound is a molecule or ion that can react both an acid as well as a base to produce salt and water.

d. The device used for grinding an ore.

Ans: The device used for grinding an ore is grinding mill.

e. The nonmetal having electrical conductivity.

Ans: Graphite an allotrope of carbon is a good conductor of electricity.

f. The reagent that dissolves noble metals.

Ans: Aqua regia is 1:3 mixture of concentrated nitric and hydrochloric acids. It dissolves noble metals such as gold, palladium, and platinum.

2. Make pairs of substances and their properties.

Answer :

Substance

a. Potassium bromide

b. Gold

c. Sulphur

d. Neon

Property

1. Soluble in water

2. High ductility

3. Combustible

4. No chemical reaction

3. Identify the pairs of metals and their ores from the following.

Answer :

Group A

a. Bauxite

b. Cassiterite

c. Cinnabar

Group B

i. Aluminium

ii. Tin

iii. Mercury

4. Explain the terms.

a. **Metallurgy** :The science and technology regarding the extraction of metals from ores and their purification for the use is called metallurgy.

The major steps involved for the extraction of a metal from its ore are:

(i) Concentration of ores (or enrichment of ore)

(ii) Conversion of concentrated ore into metal

(iii) Refining (purification) of impure metal

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