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NATIONAL CERTIFICATE
FINANCIAL ACCOUNTING N6
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28 November 2016 (X-Paper)
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This question paper consists of 11 pages and answer book of 11 pages.

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web the pcl 3 lewis structure has the typical case of phosphorus p in the center with 3 bonds to 3 other atoms phosphorus is from the same column as nitrogen in the periodic table meaning that p and n generally have the same bonding structure note the lone pair dots without bonds on top of p just like for n in the previous example for nh 3 chlorine cl

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web this exercise required us to take the formulas in the data table and calculate the number of valence electrons using the
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