Use the following equation to calculate the PERCENT WEIGHT CHANGE Timed weight - Initial weight x 100 - % weight change

Table 2: Diffusion & Osmosis - Effect of Temperature - Percent Weight C

| Temperature (°C) | O minutes | 5 minutes | Timed Weight 10 minutes | Thread Weight | Timed Weight 20 minutes | - | Water |
|----------------------|-----------|------------|-------------------------|---------------|-----------------------------|-----------------------|--------|
| Warm water (40°C) | 0 | 15-8-14.4 | 16-5-14-4 | 17.5-14.4 | 18.6.144 | STATE OF THE PARTY OF | 25 min |
| Room temp water | 0 | 14.3-13.1 | 14.60% 14.60% | 15.5-13.1 | 1601 - 1301 | 84.0390 | Vello |
| ce water | 0 | 16-3 - K-4 | | 18.32 % | 13.1 23.66% 17.3.15.4 | 26.71% | NOAT |
| | | 9.09 3/0 | 111696 | 11.68% | | 17.53% | Wests |

Part Two: Diffusion and Osmosis - The Effect of Concentration

- 19. Read through the experiment. Construct a hypothesis about the relationship between concentration and the rate of diffusion and osmosis. Include all concentrations in your hypothesis statement(s). The greater the difference between the concentration of water inside a cell and the concentration outside a cell, the faster the rate of asmosis.
- 20. Pre-soak three pieces of the dialysis tubing. Put the dialysis tubing in a beaker, add enough dH-O to cover, and soak for five minutes.
- 21. Remove the dialysis tubing from the water. Fold one end of the bag and secure it with a plastic clamp
- 22. Using a pipette, fill each bag one third full with molasses. Leave some empty space at the top of the bag. fold over the top, and secure it with another clamp. Rinse each bag and pat dry
- One bag will contain a 25% concentration of molasses.
- A second bag will contain a 50% solution of molasses.
- 25. The third bag will contain a 100% solution of molasses.

Diffusion And Osmosis Lab Answer Key

Huangqi Zhang

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