

4. Complete the following table for the two experiments in Model 1.

Temp	Experiment A	Experiment B
Independent Variable	# of particles	Temperature
Dependent Variable	Internal Pressure	Internal Pressure
Controlled Variable(s)	External Pressure / Volume / Temperature	External Pressure / Volume / # of particles

5. Of the variables that were controlled in both Experiment A and Experiment B in Model 1, one requires a nonflexible container. Name this variable, and explain why a nonflexible container is necessary. In your answer, consider the external and internal pressure data given in Model 1.

Experiment Requires a nonflexible container as the independent variable is # of particles which would normally expand a container. The second experiment does not require a nonflexible container as the # of particles does not change. \uparrow temp \neq \uparrow Volume

Read This!

Pressure is caused by molecules hitting the sides of a container or other objects. The pressure changes when the molecules change *how often* or *how hard* they hit. A nonflexible container is needed if the gas sample is going to have an internal pressure that is different from the external pressure. If a flexible container is used, the internal pressure and external pressure will always be the same because they are both

Gas Variables By Pogil Answers

C. E. R. Bruce



Gas Variables By Pogil Answers:

Science Citation Index, 1992 Vols for 1964 have guides and journal lists
C. E. R. Bruce, 1957

Gas Movements in the Long-period Variables

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