

Unit Test 1: Matter, Chemical Trends, & Chemical Bonding**Multiple Choice (10 marks)** (KU: 3, C: 3, I: 2)

Identify the choice that best completes the statement or answers the question. Place your answers in the key below.

1. _____ 2. _____ 3. _____ 4. _____ 5. _____
6. _____ 7. _____ 8. _____ 9. _____ 10. _____

- The element found in the periodic table in Group 6 and Period 5 is
a. tungsten, W c. tantalum, Ta
b. molybdenum, Mo d. tellurium, Te
- Which of the following groups is known for having both metallic and non-metallic properties?
a. alkali metals c. transition metals
b. alkali earth metals d. metalloids
- How many electrons does a strontium atom have in its outer shell?
a. 1 c. 7
b. 2 d. 8
- Which particles are responsible for the similar chemical and physical properties of elements in the same group?
a. protons and neutrons c. valence electrons
b. number of energy shells d. isotopes
- Which of the following chemical formulas most likely represents a pure substance containing ionic bonds?
a. Cl_2 c. CO_2
b. SO_2 d. BaO
- Consider the equation $\text{X}_{\text{g}} + e^- \rightarrow \text{X}^- + \text{energy}$. The "energy" term in the equation represents
a. electronegativity c. ionization energy
b. electron affinity d. nuclear charge
- Elements A, B, C, and D (found in Groups 1–17) have atomic radii of $265 \mu\text{m}$, $160 \mu\text{m}$, $185 \mu\text{m}$, and $175 \mu\text{m}$, respectively. Which element will most likely have the lowest ionization energy?
a. A c. C
b. B d. D
- List the following elements from smallest to largest atomic size:
I. barium II. strontium III. bismuth IV. iodine
a. III, I, II, IV c. IV, II, III, I
b. IV, III, II, I d. IV, II, I, III
- Which shape and bond angle is predicted by VSEPR theory for NF_3 ?
a. trigonal pyramid, less than 109.5° c. trigonal planar, 109.5°
b. tetrahedral, 109.5° d. trigonal planar, 120°

Answers To Unit 5 Post Test Chemistry

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Answers To Unit 5 Post Test Chemistry:

Unveiling the Magic of Words: A Report on "**Answers To Unit 5 Post Test Chemistry**"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Answers To Unit 5 Post Test Chemistry**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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