PAPER 1

Section A.

For each question there are four possible answers, A, B, C, and D. Choose the one you consider to be correct.

1 Naturally occurring silicon is a mixture of three isotopes, ²⁸Si, ²⁹Si and ³⁰Si. The relative atomic mass of silicon is 28.109.

What could be the relative abundance of each of the three isotopes?

- A 91.1% 28Si, 7.9% 29Si and 1.0% 30Si
- B 92.2% 25 Si, 4.7% 29 Si and 3.1% 30 Si
- C 95.0 % ²⁶Si, 3.2 % ²⁹Si and 1.8 % ³⁰Si
- D 96.3 % ²⁶Si, 0.3 % ²⁵Si and 3.4 % ³⁶Si

(2007 P1 A1)

2 When iron is reacted with aqueous iron(III) ions, iron(II) ions are formed.

Assuming the reaction goes to completion, how many moles of Fe and of Fe^{3*}(aq) would result in a mixture containing equal numbers of moles of Fe^{3*}(aq) and Fe^{2*}(aq) once the reaction had taken place?

	moles of Fe	moles of Fe ³ *(aq)
A	1	2
в	1	3
C	1	5
D	2	3

(2007 P1 A2)

3 Use of the Data Booklet is relevant to this question.

How many molecules are present in 1cm3 of oxygen gas under room conditions?

(2008 P1 A1)

Answers H1 Chemistry Paper

LP Steffe

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structure of chemistry teaching in numerous countries more transparent and to facilitate communication between these countries Especially in the case of the school subject chemistry which is very unpopular on the one hand and occupies an exceptional position on the other hand due to its relevance to jobs and everyday life and most notably due to its importance for innovation capacity and problem solving we have to learn from each others educational systems **Nuclear Science**Abstracts ,1975-04 Energy Research Abstracts ,1992-05 Journal of Physical & Colloid Chemistry Wilder

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Table of Contents Answers H1 Chemistry Paper

- 1. Understanding the eBook Answers H1 Chemistry Paper
 - The Rise of Digital Reading Answers H1 Chemistry Paper
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Answers H1 Chemistry Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - ∘ Features to Look for in an Answers H1 Chemistry Paper
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Answers H1 Chemistry Paper
 - Personalized Recommendations

- Answers H1 Chemistry Paper User Reviews and Ratings
- Answers H1 Chemistry Paper and Bestseller Lists
- 5. Accessing Answers H1 Chemistry Paper Free and Paid eBooks
 - Answers H1 Chemistry Paper Public Domain eBooks
 - Answers H1 Chemistry Paper eBook Subscription Services
 - Answers H1 Chemistry Paper Budget-Friendly Options
- 6. Navigating Answers H1 Chemistry Paper eBook Formats
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 - Answers H1 Chemistry Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers H1 Chemistry Paper
 - Highlighting and Note-Taking Answers H1 Chemistry Paper
 - Interactive Elements Answers H1 Chemistry Paper
- 8. Staying Engaged with Answers H1 Chemistry Paper
 - o Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers H1 Chemistry Paper
- 9. Balancing eBooks and Physical Books Answers H1 Chemistry Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers H1 Chemistry Paper
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answers H1 Chemistry Paper
 - Setting Reading Goals Answers H1 Chemistry Paper
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answers H1 Chemistry Paper
 - Fact-Checking eBook Content of Answers H1 Chemistry Paper

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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