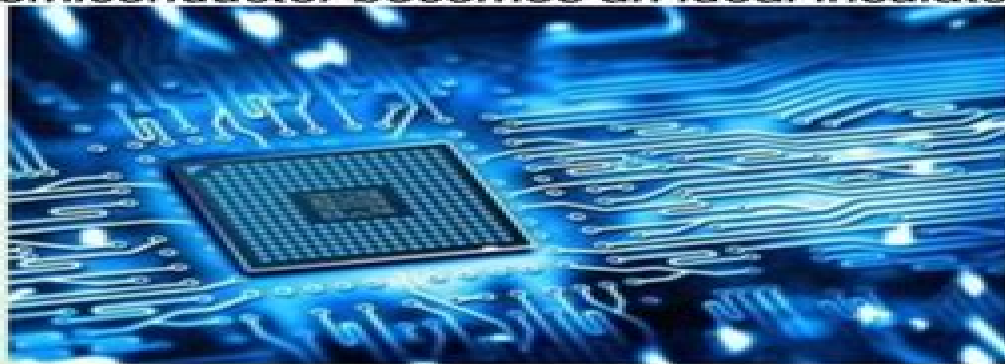


INTRODUCTION

Semiconductors :-

Most of the solids can be placed in one of the two classes: Metals and insulators. Metals are those through which electric charge can easily flow, while insulators are those through which electric charge is difficult to flow. This distinction between the metals and the insulators can be explained on the basis of the number of free electrons in them. Metals have a large number of free electrons which act as charge carriers, while insulators have practically no free electrons.

There are however, certain solids whose electrical conductivity is intermediate between metals and insulators. They are called 'Semiconductors'. Carbon, silicon and germanium are examples of semiconductors. In semiconductors the outer most electrons are neither so rigidly bound with the atom as in an insulator, nor so loosely bound as in metal. At absolute zero a semiconductor becomes an ideal insulator.



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