

PHYSICAL SCIENCES - GRADE 10

1 Revision of matter (Grade 9)

1.1 Matter and classification

- Matter is made up of particles (atoms or molecules) and it is the properties of the atoms or molecules that determine the characteristics and reactivity of that matter.
- The properties of matter include: strength; density; melting and boiling points; whether the material is brittle, malleable or ductile; whether it is magnetic or not; and its electrical and thermal conductivity.
- Elements and compounds are classified as pure because they contain particles that are all the same. Substances can be classified as pure – elements and compounds
 - or mixtures. Pure substances contain particles that are all the same. Elements and compounds can be represented by symbols and formulae. Some of the names, symbols or formulae of a number of common elements and compounds that you should have learnt are the following:
 - Elements: S for sulphur; C for carbon; P for phosphorus (note the spelling); H for hydrogen; O for oxygen; He for helium; N for nitrogen; F for fluorine; Mg for magnesium; Ca for calcium; Zn for zinc.
 - Compounds: H_2O for water; H_2SO_4 for sulphuric acid; HCl for hydrochloric acid; NaCl for sodium chloride (table salt); $CaCO_3$ for calcium carbonate; KNO_3 for potassium nitrate; Na_3PO_4 for sodium phosphate.
- You will notice that the naming of compounds follows these rules
 - The metal or hydrogen is always named first, then the non-metal (potassium iodide: KI; hydrogen chloride: HCl (also called hydrochloric acid)).
 - The name of a non-metal ion of an element always ends in -ide (for instance, sodium chloride: NaCl).
 - The name of a non-metal complex ion (more than one element in the ion) usually ends in -ate or -ite (for instance, copper sulphate: $CuSO_4$; potassium sulphite: K_2SO_3). An exception to this is hydroxide (for instance, sodium hydroxide: NaOH).
- Mixtures consist of different substances, and can be homogeneous (have the same 'look' throughout) or heterogeneous (you can see the different substances). They can be separated in various ways, for instance filtering, sieving, chromatography, distilling or dissolving one constituent.
- Some of the properties of metals are that they conduct electricity and heat, and they are malleable and ductile.
- Non-metals are non-conductors of electricity and heat. They are gases, liquids or brittle solids, and bond with each other.
- Metalloids are elements on the border between metals and non-metals, and they have both metal and non-metal properties.

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Selma Kurra



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Sound and Light Michael J. Padilla,2002 **Resources in Education** ,1997 **Science Made Easy: Introduction** Thomas Twining,1876 *Focus on physical science* Charles H. Heimler,1989 Educational Times ,1889 **Chemical News and Journal of Physical Science** ,1898 **The American** ,1899 *Bookseller* ,1871 Vols for 1871 76 1913 14 include an extra number The Christmas bookseller separately paged and not included in the consecutive numbering of the regular series *The Conservator* ,1899 **Education Outlook** ,1892 *"The" Athenaeum* ,1844 Assessment of Mars Science and Mission Priorities National Research Council,Division on Engineering and Physical Sciences,Space Studies Board,Committee on Planetary and Lunar Exploration,2003-07-08 Within the Office of Space Science of the National Aeronautics and Space Administration NASA special importance is attached to exploration of the planet Mars because it is the most like Earth of the planets in the solar system and the place where the first detection of extraterrestrial life seems

most likely to be made The failures in 1999 of two NASA missions Mars Climate Orbiter and Mars Polar Lander caused the space agency's program of Mars exploration to be systematically rethought both technologically and scientifically A new Mars Exploration Program plan summarized in Appendix A was announced in October 2000 The Committee on Planetary and Lunar Exploration COMPLEX a standing committee of the Space Studies Board of the National Research Council was asked to examine the scientific content of this new program This goals of this report are the following Review the state of knowledge of the planet Mars with special emphasis on findings of the most recent Mars missions and related research activities Review the most important Mars research opportunities in the immediate future Review scientific priorities for the exploration of Mars identified by COMPLEX and other scientific advisory groups and their motivation and consider the degree to which recent discoveries suggest a reordering of priorities and Assess the congruence between NASA's evolving Mars Exploration Program plan and these recommended priorities and suggest any adjustments that might be warranted

The Educational Times, and Journal of the College of Preceptors ,1893 *The Academy* ,1876 *Physics for the Inquiring Mind* Eric M. Rogers,2011-04-17 In our scientific age an understanding of physics is part of a liberal education Lawyers bankers governors business heads administrators all wise educated people need a lasting understanding of physics so that they can enjoy those contacts with science and scientists that are part of our civilization both materially and intellectually They need knowledge and understanding instead of the feelings all too common that physics is dark and mysterious and that physicists are a strange people with incomprehensible interests Such a sense of understanding science and scientists can be gained neither from sermons on the beauty of science nor from the rigorous courses that colleges have offered for generations when the headache clears away it leaves little but a confused sense of mystery Nor is the need met by survey courses that offer a smorgasbord of tidbit they give science a bad name as a compendium of information or formulas The non scientist needs a course of study that enables him to learn real science and make its own with delight For lasting benefits the intelligent non scientist needs a course of study that enables him to learn genuine science carefully and then encourages him to think about it and use it He needs a carefully selected framework of topics not so many that learning becomes superficial and hurried not so few that he misses the connected nature of scientific work and thinking He must see how scientific knowledge is built up by building some scientific knowledge of his own by reading and discussing and if possible by doing experiments himself He must think his own way through some scientific arguments He must form his own opinion with guidance concerning the parts played by experiment and theory and he must be shown how to develop a taste for good theory He must see several varieties of scientific method at work And above all he must think about science for himself and enjoy that These are the things that this book encourages readers to gain by their own study and thinking *Physics for the Inquiring Mind* is a book for the inquiring mind of students in college and for other readers who want to grow in scientific wisdom who want to know what physics really is **English Mechanic and Mirror of Science and Arts** ,1867

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