

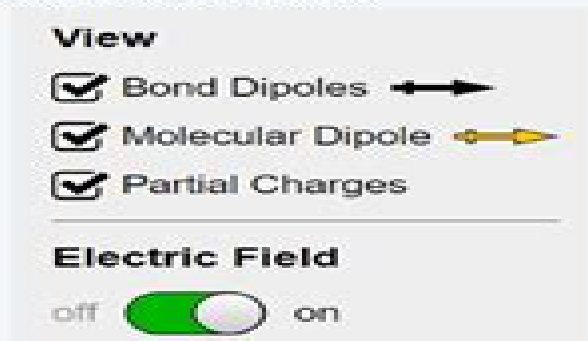
VIRTUAL WORKSHEET

MOLECULAR POLARITY

PhET SIMULATION

Go to [this link](#) and click on "Three atoms"

In the gray box, turn on all of the options, like you see below:



The yellow, green, and pink boxes at the bottom allow you to adjust the electronegativities of each atom. Before you play around with these, notice that atom B is facing the positive electric field. Why is this?

This is because atom B has a positive charge, therefore it is attracted toward the positive electric field.

Now, set atom B to the highest electronegativity. Notice the partial charges. Now, move the electronegativity of atom C to the third or fourth tick mark. What do you notice about the partial charges?

I noticed that when atom C has more electronegativity, it is drawn toward the positive electric field.

Play around with the atoms. You can also change the shape by clicking and dragging.

Click on "Real Molecules" at the bottom of the screen. A message will appear that says, "Real Molecules is not currently available in HTML5." Click the link provided in that message.

A new tab will appear. Click the *play* symbol and then click the first option "Run CheerpJ Browser-Compatible Version." This will take a minute to load. Once it does, you'll need to click the "Real Molecules" tab.

In the gray box, turn on the options I have turned on:

Phet Molecular Polarity Simulation Answers

Hongru Du

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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