

Assignment #1 – Compound Names and Formulas
Single-valent ions only

A. Name these Compounds

1. Li_2S Lithium Sulfide
2. CaO Calcium Oxide
3. NaF Sodium Fluoride
4. CaBr_2 Calcium Bromide
5. MgCl_2 Magnesium Chloride
6. BBr_3 Boron Bromide
7. Cs_2O Cesium Oxide
8. FrBr Francium Bromide
9. Ag_2S Silver Sulfide

10. GeF_4 Germanium Fluoride
11. Ga_2O_3 Gallium Oxide
12. EsCl_3 Einsteinium Chloride
13. Fm_2O_3 Fermium Oxide
14. Mg_3N_2 Magnesium Nitride
15. Rb_2O Rubidium Oxide
16. RaO Radium Oxide
17. SrO Strontium Oxide
18. Tc_2O_7 Technetium Oxide

B. Write the correct chemical formula for these compounds by balancing the ionic charges

1. sodium chloride NaCl
2. magnesium fluoride MgF_2
3. silver oxide Ag_2O
4. indium bromide InBr_3
5. zinc bromide ZnBr_2
6. neodymium oxide Nd_2O_3
7. thorium sulphide ThS_2
8. actinium oxide Ac_2O_3
9. radium bromide RaBr_2
10. cesium oxide Cs_2O

11. hydrogen oxide H_2O
12. francium nitride Fr_3N
13. rubidium phosphide Rb_3P
14. potassium oxide K_2O
15. beryllium sulphide BeS
16. lithium sulphide Li_2S
17. hydrogen bromide HBr
18. strontium nitride Sr_3N_2
19. calcium oxide CaO
20. tantalum nitride Ta_3N_5

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Ensheng Dong



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Table of Contents Flinn Scientific Ionic Formula Writing Kit Answers

1. Understanding the eBook Flinn Scientific Ionic Formula Writing Kit Answers
 - The Rise of Digital Reading Flinn Scientific Ionic Formula Writing Kit Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Flinn Scientific Ionic Formula Writing Kit Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Flinn Scientific Ionic Formula Writing Kit Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Flinn Scientific Ionic Formula Writing Kit Answers
 - Personalized Recommendations
 - Flinn Scientific Ionic Formula Writing Kit Answers User Reviews and Ratings
 - Flinn Scientific Ionic Formula Writing Kit Answers and Bestseller Lists
5. Accessing Flinn Scientific Ionic Formula Writing Kit Answers Free and Paid eBooks
 - Flinn Scientific Ionic Formula Writing Kit Answers Public Domain eBooks
 - Flinn Scientific Ionic Formula Writing Kit Answers eBook Subscription Services
 - Flinn Scientific Ionic Formula Writing Kit Answers Budget-Friendly Options
6. Navigating Flinn Scientific Ionic Formula Writing Kit Answers eBook Formats

- ePub, PDF, MOBI, and More
 - Flinn Scientific Ionic Formula Writing Kit Answers Compatibility with Devices
 - Flinn Scientific Ionic Formula Writing Kit Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Flinn Scientific Ionic Formula Writing Kit Answers
 - Highlighting and Note-Taking Flinn Scientific Ionic Formula Writing Kit Answers
 - Interactive Elements Flinn Scientific Ionic Formula Writing Kit Answers
 8. Staying Engaged with Flinn Scientific Ionic Formula Writing Kit Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Flinn Scientific Ionic Formula Writing Kit Answers
 9. Balancing eBooks and Physical Books Flinn Scientific Ionic Formula Writing Kit Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Flinn Scientific Ionic Formula Writing Kit Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Flinn Scientific Ionic Formula Writing Kit Answers
 - Setting Reading Goals Flinn Scientific Ionic Formula Writing Kit Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Flinn Scientific Ionic Formula Writing Kit Answers
 - Fact-Checking eBook Content of Flinn Scientific Ionic Formula Writing Kit Answers
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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