#### LA COMPOSIZIONE DELLA MATERIA

L'oggeto di studio della chimica è la materia. La chimica è infat la disciplina che studia:

- La strutura e la composizione della materia
- Le trasformazioni che la materia subisce
- L'energia coinvolta in queste trasformazioni

#### MISCELE E SOSTANZE

La materia è composta di atomi, partcelle piccolissime che possono combinarsi tra loro in modo diverso.

Riguardo alla composizione chimica, la materia può essere soto forma di:

- Miscele (o miscugli).
- Sostanze (o sostanze pure)

#### MISCELE

Una miscela è un insieme fisico di due o più sostanze e ha una composizione variabile. In base alla possibilità di distnguerne i component, le miscele vengono divise in:

- Eterogenee: miscele formate da due o più fasi (una fase è una porzione di un sistema in cui le proprietà
  - fisiche intensive sono identche in ogni punto, separata dal resto da superfici limitant) che non hanno
  - composizione uniforme in tut i punt del campione e che i component sono facilmente distinguibili.

Esempi di miscele eterogenee sono le rocce, il sangue e il late.

- Omogenee: miscele formate da una sola fase che hanno una composizione uniforme in tut
  i punt del
  - campione e che i component non sono distnguibili. Le miscele omogenee sono dete anche soluzioni.

Esempl di miscele omogenee sono l'acqua di mare, il vino e l'aria.

Separando una miscela nei suoi component con metodi fisici (es. filtrazione o distilazione) si otengono sostanze pure. La scelta del metodo di separazione dipende dallo stato fisico dei component e dalle loro proprietà chimiche e fisiche.

I METODI DI SEPARAZIONE maggiormente utilizzat sono:

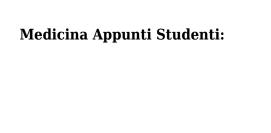
- Filtrazione: permete di separare i component di una miscela eterogenea costtuita da un liquido e un solido
  - sfrutando la diversa dimensione delle partcelle. Essa si realizza facendo passare la miscela atraverso un filtro di carta assorbente.
- Distilizzione: permete di separare i component di una miscela eterogenea sfrutandone la diversa

temperatura di ebollizione. Essa può essere:

Semplice quando si separa un liquido da una soluzione solido-liquido Frazionata quando si separa un liquido da una soluzione di due o più liquidi

# **Medicina Appunti Studenti**

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