Use the following equation to calculate the PERCENT WEIGHT CHANGE Timed weight - Initial weight x 100 - % weight change

Table 2: Diffusion & Osmosis - Effect of Temperature - Percent Weight C

Temperature (°C)	O minutes	5 minutes	Timed Weight 10 minutes	Thread Weight	Timed Weight 20 minutes	-	Water
Warm water (40°C)	0	15-8-14.4	16-5-14-4	17.5-14.4	18.6.144	STATE OF THE PARTY OF	25 min
Room temp water	0	14.3-13.1	14.60% 14.60%	15.5-13.1	1601 - 1301	84.0390	Vello
ce water	0	16-3 - K-4		18.32 %	13.1 23.66% 17.3.15.4	26.71%	NOAT
		9.09 3/0	111696	11.68%		17.53%	Wests

## Part Two: Diffusion and Osmosis - The Effect of Concentration

- 19. Read through the experiment. Construct a hypothesis about the relationship between concentration and the rate of diffusion and osmosis. Include all concentrations in your hypothesis statement(s). The greater the difference between the concentration of water inside a cell and the concentration outside a cell, the faster the rate of asmosis.
- 20. Pre-soak three pieces of the dialysis tubing. Put the dialysis tubing in a beaker, add enough dH-O to cover, and soak for five minutes.
- 21. Remove the dialysis tubing from the water. Fold one end of the bag and secure it with a plastic clamp
- 22. Using a pipette, fill each bag one third full with molasses. Leave some empty space at the top of the bag. fold over the top, and secure it with another clamp. Rinse each bag and pat dry
- One bag will contain a 25% concentration of molasses.
- A second bag will contain a 50% solution of molasses.
- 25. The third bag will contain a 100% solution of molasses.

# **Diffusion And Osmosis Lab Answer Key**

National Research Council, Division of Behavioral and Social Sciences and Education, Center for Education, Board on Science Education, Committee on High School Laboratories: Role and Vision

#### **Diffusion And Osmosis Lab Answer Key:**

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