

In Table 4 below,  $[\text{Fe}^{3+}]$  after mixing is obtained by using  $M_1V_1 = M_2V_2$ . This data is to be transferred to Table 5 below as the initial  $[\text{Fe}^{3+}]$  (column 2).

Similarly in Table 4 of Data sheet  $[\text{SCN}^-]$  after mixing obtained using  $M_1V_1 = M_2V_2$ . This data is then to be transferred to Table 5 below as the initial  $[\text{SCN}^-]$  (column 3).

Table 4. Initial Concentration Data

| Test Tube | $[\text{Fe}^{3+}]$<br>Before<br>Mixing | $[\text{Fe}^{3+}]$<br>After Mixing <sup>1</sup><br>(Initial $[\text{Fe}^{3+}]$ ) | $[\text{SCN}^-]$<br>Before<br>Mixing | $[\text{SCN}^-]$<br>After Mixing <sup>1</sup><br>(Initial $[\text{SCN}^-]$ ) |
|-----------|--|--|--------------------------------------|--|
| F         | 0.01                                   | 0.0025M  | 0.00113                              | $2.825 \times 10^{-4}$   |
| G         | 0.01                                   | 0.0025M  | 0.00113                              | $5.65 \times 10^{-4}$  |
| H         | 0.01                                   | 0.005M   | 0.00113                              | $2.825 \times 10^{-4}$   |
| I         | 0.01                                   | 0.005M   | 0.00113                              | $1.13 \times 10^{-4}$  |

(Show calculations for initial concentration data on separate sheet)

In Table 5 below, Equilibrium  $[\text{FeSCN}^{2+}]$  is obtained from the Calibration line using absorbance values of solutions F through I.

Equilibrium  $[\text{Fe}^{3+}] = \text{Initial } [\text{Fe}^{3+}] - \text{Equilibrium } [\text{FeSCN}^{2+}]$

Equilibrium  $[\text{SCN}^-] = \text{Initial } [\text{SCN}^-] - \text{Equilibrium } [\text{FeSCN}^{2+}]$

Table 5. Equilibrium Constant Calculation

| Test Tube | Initial $[\text{Fe}^{3+}]$ | Initial $[\text{SCN}^-]$ | Equilibrium $[\text{FeSCN}^{2+}]$<br>(from graph) | Equilibrium $[\text{Fe}^{3+}]$ | Equilibrium $[\text{SCN}^-]$ | $K_c$ |
|-----------|----------------------------|--------------------------|---|--------------------------------|------------------------------|-------|
| F         | 0.0025                     | $2.825 \times 10^{-4}$   |   |                                |                              |       |
| G         | 0.0025                     | $5.65 \times 10^{-4}$    |   |                                |                              |       |
| H         | 0.005                      | $2.825 \times 10^{-4}$   |   |                                |                              |       |
| I         | 0.005                      | $1.13 \times 10^{-4}$    |   |                                |                              |       |

(Show calculations for Equilibrium Constant on separate sheet)

Average Equilibrium Constant \_\_\_\_\_

# Determining Equilibrium Constant Lab Report Answers

**Zhe Xu**



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unemployment was at 25 percent and more than 5 000 banks had gone out of business

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are out of work and hungry there are not many women in the bread line