

I. PRODUCT IDENTIFICATION

Chemical Trade Name (as used on label):

Lead-Acid Battery

Manufacturer's Name/Address:

EnerSys
P.O. Box 14145
2306 Bernville Road
Reading, PA 19612-4145

Chemical Family/Classification:

Electric Storage Battery

Telephone:

For information and emergencies, contact EnerSys' Environmental, Health & Safety Dept. at 610-208-1996

24-Hour Emergency Response Contact:

CHEMTREC (DOMESTIC): 800-424-9300 CHEMTREC INTL: 703-527-3877

II. HAZARDOUS INGREDIENTS/IDENTIFY INFORMATION

			Air Exposure Limits (mg/m ³)		
Components	CAS Number	Approximate % by Wt. Or Vol.	OSHA	ACGIH	NIOSH
Inorganic Lead Compounds:					
Lead	7439-92-1	60	50	150	100
* Antimony	7440-36-0	2	500	500	—
* Arsenic	7440-38-2	0.2	10	200	—
* Cadmium	7440-70-2	0.2	—	—	—
* Tin	7440-51-5	0.2	2000	2000	—
Electrolyte (Sulfuric Acid)	7664-93-9	19-30	1000	1000	1000
Case Materials:					
Polystyrene	9003-07-0	5-10	N/A	N/A	N/A
Polystyrene	9003-53-6				
Styrene Acrylonitrile	9003-54-7				
Acrylonitrile Butadiene Styrene	9003-56-9				
Styrene Butadiene	9003-55-8				
Polyvinylchloride	9002-86-2				
Polycarbonate, Hard Rubber, Polyethylene	—				
Other:					
Silicon Dioxide (Gel batteries only)	7631-86-9	20-40	N/A	N/A	N/A
Sheet Molding Compound	—		N/A	N/A	N/A
Glass reinforced polyester					

Inorganic lead and electrolyte (sulfuric acid) are the primary components of every battery manufactured by EnerSys.

Other ingredients may be present dependent upon battery type. Contact your EnerSys representative for additional information.

III. PHYSICAL DATA

Electrolyte:			
Boiling Point:	203 - 240° F	Specific Gravity (H ₂ O = 1):	1.215 to 1.350
Melting Point:	N/A	Vapor Pressure (mm Hg):	10
Solubility in Water:	100%	Vapor Density (AIR = 1):	Greater than 1
Evaporation Rate: (Butyl Acetate = 1)	Less than 1	% Volatile by Weight:	N/A
Appearance and Odor:	Manufactured article; no apparent odor. Electrolyte is a clear liquid with a sharp, penetrating, pungent odor.		

IV. FIRE AND EXPLOSION HAZARD DATA

Flash Point: N/A	Flammable Limits: LEL = 4.1% (Hydrogen Gas)	UEL = 74.2%
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Extinguishing Media: CO₂; foam; dry chemical

Special Fire Fighting Procedures:

If batteries are on charge, shut off power. Use positive pressure, self-contained breathing apparatus. Water applied to electrolyte generates heat and causes it to spatter. Wear acid-resistant clothing.

Unusual Fire and Explosion Hazards:

Highly flammable hydrogen gas is generated during charging and operation of batteries. To avoid risk of fire or explosion, keep sparks or other sources of ignition away from batteries. Do not allow metallic materials to simultaneously contact negative and positive terminals of cells and batteries. Follow manufacturer's instructions for installation and service.

V. REACTIVITY DATA

Stability: Stable

Conditions To Avoid: Prolonged overcharge; sources of ignition

Incompatibility: (Materials to avoid)

Sulfuric Acid: Contact with combustibles and organic materials may cause fire and explosion. Also reacts violently with strong reducing agents, metals, sulfur trioxide gas, strong oxidizers and water. Contact with metals may produce toxic sulfur dioxide fumes and may release flammable hydrogen gas.

Lead Compounds: Avoid contact with strong acids, bases, halides, halogenates, potassium nitrate, permanganate, peroxides, nascent hydrogen and reducing agents.

Hazardous Decomposition Products:

Sulfuric Acid: Sulfur trioxide, carbon monoxide, sulfuric acid mist, sulfur dioxide, and hydrogen.

Lead Compounds: High temperatures likely to produce toxic metal fume, vapor, or dust; contact with strong acid or base or presence of nascent hydrogen may generate highly toxic arsine gas.

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DP Hallahan

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