

AP Chemistry: Chapter 5-6 Student Notes

Objectives

Chapter 5: Gases

- ☐ Kinetic Molecular Theory of Gases
- ☐ Understand pressure and its units
- ☐ Boyles-Charles-Gay-Lussac-Avogadro's Laws (Conceptual and Calculations)
- ☐ Ideal Gas Law
- ☐ Gas Stoichiometry
- ☐ Daltons Law of Partial Pressures (and applications to Gas Stoichiometry)
- ☐ Grahams Law
- ☐ Real verses Ideal Gases

Chapter 6: Thermochemistry

- ☐ The First Law of Thermodynamics
- ☐ Energy in a chemical reaction (exothermic and endothermic)
- ☐ Enthalpy ΔH (4 ways to calculate it)
- ☐ Finding ΔH using calorimetry
- ☐ Hess' Law version 1: Adding reactions
- ☐ Hess' Law version 2: $\sum \text{Products} - \sum \text{Reactants}$
- ☐ Different kinds of ΔH

Kinetic Molecular Theory of Gases

Kinetic = _____

Molecular = _____

So the KMT is the _____ theory

All molecules _____

Solids move _____

Liquids _____

Gases move in _____ - _____ motion

How fast do they move?

Use $KE = \frac{1}{2}mv^2$: Solve for velocity

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CliffsNotes AP Chemistry Angela Woodward Spangenberg, 2016-01-12 Test prep for the AP Chemistry exam with 100% brand new content that reflects recent exam changes Addressing the major overhaul that the College Board recently made to the AP Chemistry exam this AP Chemistry test prep guide includes completely brand new content tailored to the exam administered every May Features of the guide include review sections of the six big ideas that the new exam focuses on Fundamental building blocks Molecules and interactions Chemical reactions Reaction rates Thermodynamics Chemical equilibrium Every section includes review questions and answers Also included in the guide are two full length practice tests as well as a math review section and sixteen discrete laboratory exercises to prepare AP Chemistry students for the required laboratory experiments section on the exam

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Table of Contents Ap Chemistry Chapter 5 6 Student Notes

1. Understanding the eBook Ap Chemistry Chapter 5 6 Student Notes
 - The Rise of Digital Reading Ap Chemistry Chapter 5 6 Student Notes
 - Advantages of eBooks Over Traditional Books
2. Identifying Ap Chemistry Chapter 5 6 Student Notes
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ap Chemistry Chapter 5 6 Student Notes
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ap Chemistry Chapter 5 6 Student Notes
 - Personalized Recommendations
 - Ap Chemistry Chapter 5 6 Student Notes User Reviews and Ratings
 - Ap Chemistry Chapter 5 6 Student Notes and Bestseller Lists
5. Accessing Ap Chemistry Chapter 5 6 Student Notes Free and Paid eBooks

- Ap Chemistry Chapter 5 6 Student Notes Public Domain eBooks
 - Ap Chemistry Chapter 5 6 Student Notes eBook Subscription Services
 - Ap Chemistry Chapter 5 6 Student Notes Budget-Friendly Options
6. Navigating Ap Chemistry Chapter 5 6 Student Notes eBook Formats
- ePub, PDF, MOBI, and More
 - Ap Chemistry Chapter 5 6 Student Notes Compatibility with Devices
 - Ap Chemistry Chapter 5 6 Student Notes Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Ap Chemistry Chapter 5 6 Student Notes
 - Highlighting and Note-Taking Ap Chemistry Chapter 5 6 Student Notes
 - Interactive Elements Ap Chemistry Chapter 5 6 Student Notes
8. Staying Engaged with Ap Chemistry Chapter 5 6 Student Notes
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ap Chemistry Chapter 5 6 Student Notes
9. Balancing eBooks and Physical Books Ap Chemistry Chapter 5 6 Student Notes
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ap Chemistry Chapter 5 6 Student Notes
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ap Chemistry Chapter 5 6 Student Notes
- Setting Reading Goals Ap Chemistry Chapter 5 6 Student Notes
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ap Chemistry Chapter 5 6 Student Notes
- Fact-Checking eBook Content of Ap Chemistry Chapter 5 6 Student Notes
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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