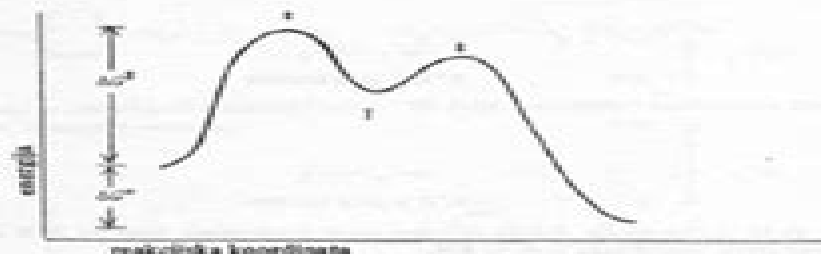


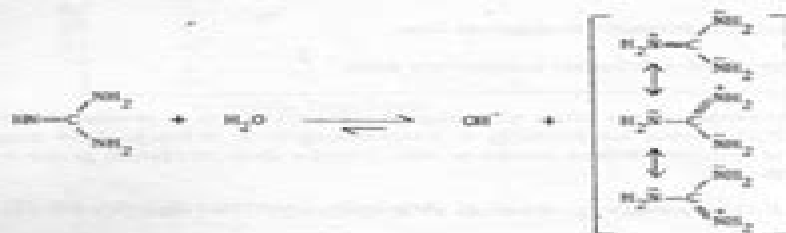
5.13

5.48



7.38

6.49. Konjugirana kiseline guanidina ima tri energetski ekvivalentne rezonantne strukture i ravnoteža je pomaknuta daleko udesno.



7.39

5.50



7.43

6.51. Dužikov atom u jednostavnim amidima kakav je spoj A nije bazidan jer zbog susjedstva karbonylne grupe dolazi do delokalizacije slobodnog elektronskog para. U spoju B takva bi konjugacija dovela do nastajanja (djelomično) dvostrukog veza na mjestu premošćenja u bicykličkom spoju. To bi bilo energetski vrlo nepovoljno (Bredtova pravila).

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