

# **PHYSICAL SCIENCES - GRADE 10**

## **1 Revision of matter (Grade 9)**

### **1.1 Matter and classification**

- Matter is made up of particles (atoms or molecules) and it is the properties of the atoms or molecules that determine the characteristics and reactivity of that matter.
- The properties of matter include: strength; density; melting and boiling points; whether the material is brittle, malleable or ductile; whether it is magnetic or not; and its electrical and thermal conductivity.
- Elements and compounds are classified as pure because they contain particles that are all the same. Substances can be classified as pure – elements and compounds
  - or mixtures. Pure substances contain particles that are all the same. Elements and compounds can be represented by symbols and formulae. Some of the names, symbols or formulae of a number of common elements and compounds that you should have learnt are the following:
    - Elements: S for sulphur; C for carbon; P for phosphorus (note the spelling); H for hydrogen; O for oxygen; He for helium; N for nitrogen; F for fluorine; Mg for magnesium; Ca for calcium; Zn for zinc.
    - Compounds:  $H_2O$  for water;  $H_2SO_4$  for sulphuric acid; HCl for hydrochloric acid; NaCl for sodium chloride (table salt);  $CaCO_3$  for calcium carbonate;  $KNO_3$  for potassium nitrate;  $Na_3PO_4$  for sodium phosphate.
- You will notice that the naming of compounds follows these rules
  - The metal or hydrogen is always named first, then the non-metal (potassium iodide: KI; hydrogen chloride: HCl (also called hydrochloric acid)).
  - The name of a non-metal ion of an element always ends in -ide (for instance, sodium chloride: NaCl).
  - The name of a non-metal complex ion (more than one element in the ion) usually ends in -ate or -ite (for instance, copper sulphate:  $CuSO_4$ ; potassium sulphite:  $K_2SO_3$ ). An exception to this is hydroxide (for instance, sodium hydroxide: NaOH).
- Mixtures consist of different substances, and can be homogeneous (have the same 'look' throughout) or heterogeneous (you can see the different substances). They can be separated in various ways, for instance filtering, sieving, chromatography, distilling or dissolving one constituent.
- Some of the properties of metals are that they conduct electricity and heat, and they are malleable and ductile.
- Non-metals are non-conductors of electricity and heat. They are gases, liquids or brittle solids, and bond with each other.
- Metalloids are elements on the border between metals and non-metals, and they have both metal and non-metal properties.

# Physical Science Chapter 10 Sound Notes Section 1 The

**M Mosston**



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Sound and Light Michael J. Padilla,2002      **Resources in Education** ,1997      **Science Made Easy: Introduction** Thomas Twining,1876      *Focus on physical science* Charles H. Heimler,1989      Educational Times ,1889      **Chemical News and Journal of Physical Science** ,1898      **The American** ,1899      *Bookseller* ,1871 Vols for 1871 76 1913 14 include an extra number The Christmas bookseller separately paged and not included in the consecutive numbering of the regular series      *The Conservator* ,1899      **Education Outlook** ,1892      *"The" Athenaeum* ,1844      Assessment of Mars Science and Mission Priorities National Research Council,Division on Engineering and Physical Sciences,Space Studies Board,Committee on Planetary and Lunar Exploration,2003-07-08 Within the Office of Space Science of the National Aeronautics and Space Administration NASA special importance is attached to exploration of the planet Mars because it is the most like Earth of the planets in the solar system and the place where the first detection of extraterrestrial life seems

most likely to be made The failures in 1999 of two NASA missions Mars Climate Orbiter and Mars Polar Lander caused the space agency's program of Mars exploration to be systematically rethought both technologically and scientifically A new Mars Exploration Program plan summarized in Appendix A was announced in October 2000 The Committee on Planetary and Lunar Exploration COMPLEX a standing committee of the Space Studies Board of the National Research Council was asked to examine the scientific content of this new program This goals of this report are the following Review the state of knowledge of the planet Mars with special emphasis on findings of the most recent Mars missions and related research activities Review the most important Mars research opportunities in the immediate future Review scientific priorities for the exploration of Mars identified by COMPLEX and other scientific advisory groups and their motivation and consider the degree to which recent discoveries suggest a reordering of priorities and Assess the congruence between NASA's evolving Mars Exploration Program plan and these recommended priorities and suggest any adjustments that might be warranted

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