

PS Chemistry Semester Test Review

Review Schedule:

- Monday, December 16: Assign Review Packet, Hand in Text Books, Clean out Binders
- Tuesday, December 17: Work on Review Packet, Hand in Text Books, Clean out Binders
- Wednesday, December 18: Correct Review Packet, Review Game, Create Legal Cheat Sheet
- Thursday, December 19: PS Chemistry Semester Test

The Test:

- The test is worth 14% of your semester grade.
- Test will be taken through BlackBoard Learn.
- 100 multiple choice questions.
- A basic calculator is recommended.
- I will provide a Periodic Table and an Oxidation Chart during the test.
- Your book must be returned to me prior to taking the test.
- Water bottle is optional.
- Bring a book to read or something to keep yourself occupied (just in case you complete the test early).
- You will not be allowed to leave the classroom during the testing period.

General Overview:

In order for the review to be most helpful, it is necessary for you to work through the review questions and come to class prepared to ask questions for clarification and review. In addition, you should look over chapter reviews and notes to help study! You can also make arrangements to look over your old tests and quizzes.

The Review:

The review is divided by chapters (or units). Below you will find a general list of terms/concepts from each unit followed by review questions. Also notice the number in the box...that is the number of test questions from the section.

Chapter 1: The Nature of Science

8

scientific method
hypothesis

constant
control

variables
standards of measurement

SI units
metric conversion

- What tools and units are used in science for measuring length, mass, volume, and density?
- A bag of apples weighs 3249 grams. What is this weight in kilograms?
- List the steps of the scientific method.
- Krissy thinks that Skin-So-Soft by Avon will repel mosquitoes. She takes 100 mosquitoes and puts them in a container, half the container contains a cloth soaked in skin-so-soft. After one hour she made observations, she found that 90 mosquitoes were on the half of the container without the rag and 10 were found near the rag. Krissy also put 100 mosquitoes into another container, half of this container contained a rag soaked in water. She observed this container an hour later and found the mosquitoes were evenly distributed across the container.
 - Write a hypothesis for Krissy.
 - What was the control group?
 - What was the independent variable?
 - What was the dependent variable?
 - What should Krissy conclude?
- Which axes are the independent and dependent variables found?

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Ryan C Fortenberry



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Questions Chapter 28 Rotational Motion Questions Chapter 29 Second Law of Thermodynamics Questions Chapter 30 Simple Harmonic Motion Questions Chapter 31 Special Relativity Questions Chapter 32 Straight Line Motion Questions Chapter 33 Transverse Waves Questions Chapter 34 Two and Three Dimensional Motion Questions Chapter 35 Vector Quantities Questions Chapter 36 Work Kinetic Energy Theorem Questions The Alternating Fields and Currents Quiz Questions PDF e Book Chapter 1 interview questions and answers on Alternating current damped oscillations in an RLS circuit electrical mechanical analog forced and free oscillations LC oscillations phase relations for alternating currents and voltages power in alternating current circuits transformers The Astronomical Data Quiz Questions PDF e Book Chapter 2 interview questions and answers on Aphelion distance from earth eccentricity of orbit equatorial diameter of planets escape velocity of planets gravitational acceleration of planets inclination of orbit to earth s orbit inclination of planet axis to orbit mean distance from sun to planets moons of planets orbital speed of planets perihelion period of rotation of planets planet densities planets masses sun earth and moon The Capacitors and Capacitance Quiz Questions PDF e Book Chapter 3 interview questions and answers on Capacitor in parallel and in series capacitor with dielectric charging a capacitor cylindrical capacitor parallel plate capacitor The Circuit Theory Quiz Questions PDF e Book Chapter 4 interview questions and answers on Loop and junction rule power series and parallel resistances single loop circuits work energy and EMF The Conservation of Energy Quiz Questions PDF e Book Chapter 5 interview questions and answers on Center of mass and momentum collision and impulse collisions in one dimension conservation of linear momentum conservation of mechanical energy linear momentum and Newton s second law momentum and kinetic energy in collisions Newton s second law for a system of particles path independence of conservative forces work and potential energy The Coulomb s Law Quiz Questions PDF e Book Chapter 6 interview questions and answers on Charge is conserved charge is quantized conductors and insulators and electric charge The Current Produced Magnetic Field Quiz Questions PDF e Book Chapter 7 interview questions and answers on Ampere s law and law of Biot Savart The Electric Potential Energy Quiz Questions PDF e Book Chapter 8 interview questions and answers on Introduction to electric potential energy electric potential and equipotential surfaces The Equilibrium Indeterminate Structures Quiz Questions PDF e Book Chapter 9 interview questions and answers on Center of gravity density of selected materials of engineering interest elasticity equilibrium indeterminate structures ultimate and yield strength of selected materials of engineering interest and Young s modulus of selected materials of engineering interest The Finding Electric Field Quiz Questions PDF e Book Chapter 10 interview questions and answers on Electric field electric field due to continuous charge distribution electric field lines flux and Gauss law The First Law of Thermodynamics Quiz Questions PDF e Book Chapter 11 interview questions and answers on Absorption of heat by solids and liquids Celsius and Fahrenheit scales coefficients of thermal expansion first law of thermodynamics heat of fusion of common substances heat of transformation heat of vaporization of common substances introduction to thermodynamics molar specific heat substance specific heat in

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