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DESCRIBING CHEMICAL REACTIONS Section Review - Objectives

- Explain how to write a word equation
- Describe how to write a skeleton equation
- List the steps for writing a complete chemical equation

Vocabulary

- chemical equation
- skeleton equation
- cutallyst.

- coefficients
- balanced equation

Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

| A chemical reaction can be concisely represented by a chemical | 1. |
|--|----|
| The substances that undergo a chemical change are the The new | 2 |
| substances formed in a chemical reaction are the3 In accordance with | 3 |
| the law of conservation of4, a chemical equation must be balanced. | 4 |
| When balancing an equation, you place5 in front of reactants and | 5 |
| products so that the same number of atoms of each 6 are on each side of | 6 |
| the equation. An equation must never be balanced by changing the7 in | 7 |
| the chemical formula of a substance. | 8 |
| Special symbols are used to show the physical state of a substance in a | 9. |
| reaction. The symbol for a liquid is8; for a solid,9; and for a gas, | 10 |
| 10 A substance dissolved in water is designated11 If a12 is | 11 |
| used to increase the rate of a chemical reaction, its formula is written above the | 12 |
| | |

Part B True-False

autocolec.

Classify each of these statements as always true, AT; sometimes true, ST; or never true, NT.

13. In an equation, a substance is shown to be in the gaseous state by placing an upward-pointing arrow after its formula.

14. The symbol Δ placed over the arrow in an equation means that heat is supplied to the reaction.

15. Atoms are destroyed in a chemical reaction.

16. A skeleton equation is not a balanced equation.

Part D Questions and Problems

24. Write a balanced equation for each of these chemical reactions. I
a. Aluminum reacts with aqueous hydrochloric acid to form hydrogen gas and aqueous aluminum chloride.

b. Acetylene gas (C₂H₂) burns in a welding torch with oxygen to form carbon dioxide gas and water vapor.

Describing Chemical Reactions Worksheet Answers

Victor G. Ganzha, Ernst W. Mayr, Evgenii V. Vorozhtsov

Describing Chemical Reactions Worksheet Answers:

Chemical Misconceptions Keith Taber, 2002 Part 2 provides strategies for dealing with some of the misconceptions that students have by including ready to use classroom resources Cambridge IGCSETM Chemistry Teacher's Guide (Collins Cambridge IGCSETM) Chris Sunley, 2022-02-03 Prepare students with complete coverage of the revised Cambridge IGCSETM Chemistry syllabus 0620 0971 for examination from 2023 Collins Cambridge IGCSE Chemistry Teacher's Guide is full of lesson ideas practical instructions technician s notes planning support and more **Learning Elementary** Chemistry for Class 7 (A.Y. 2023-24)Onward Dr. R. Goel, 2023-05-20 The series Learning Elementary Chemistry for Classes 6 to 8 has been revised strictly according to the latest curriculum. The content of this series has been developed to fulfill the requirement of all the six domains Concepts Processes Applications Attitudes Creativity and World view of Science to make teaching and learning of Chemistry interesting understandable and enjoyable for young minds This series builds a solid foundation for young learners to prepare them for higher classes The main strength of the series lies in the subject matter and the experience that a learner will get in solving difficult and complex problems of Chemistry Emphasis has been laid upon mastering the fundamental principles of Chemistry rather than specific procedures Unique features of this series are The content of the book is written in a very simple and easy to understand language All the Key concepts in the curriculum have been systematically covered and graded in the text Each theme has been divided into units followed by thought provoking and engaging exercises to test the knowledge understanding and applications of the concepts learnt in that unit At the end of each theme a comprehensive theme assignment which is aligned with the guidelines provided in National Education Policy NEP 2020 is given Explanations illustrations diagrams experiments and solutions to numerical problems have been included to make the subject more interesting comprehensive and appealing Diagrams illustrations and text have been integrated to enhance comprehension Definitions and other important scientific information are highlighted Throughout the series investigations related to the text enable the learners to learn through experimentation Quick revision of each chapter has been given under the caption Highlights in Review Online Support It provides Video lectures Unit wise interactive exercises Chapterwise Worksheet Solution of textbook questions for Teachers only E Book for Teachers only I hope this series would meet the needs and requirements of the curriculum to achieve the learning outcomes as laid down in the curriculum Suggestions and constructive feedback for the further improvement of the book shall be gratefully acknowledged and incorporated in the future edition of the book Author Holt Chemistry Holt Rinehart & Winston, 2003-01-24 Prentice Hall Science Explorer Michael J. Padilla, 2002 **CBSE** Chapterwise Worksheets for Class 9 Gurukul, 2021-07-30 Practice Perfectly and Enhance Your CBSE Class 9th preparation with Gurukul s CBSE Chapterwise Worksheets for 2022 Examinations Our Practicebook is categorized chapterwise topicwise to provide you in depth knowledge of different concept topics and questions based on their weightage to help you perform better in the 2022 Examinations How

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approaches from CER literature and beyond This Research Topic bridges discipline based education research and the scholarship of teaching and learning which will help advance organic chemistry education and improve student outcomes

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