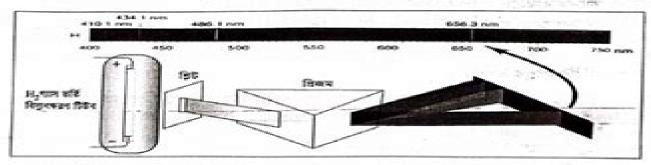
# গুণগত রসায়ন Qualitative Chemistry

#### ভূমিকা (Introduction)

এ বিশ্ব প্রকৃতি বিজ্ঞানীদের কাছে দু'ভাগে বিভক্ত, যেমন পদার্থ (matter) ও বিকীর্থ শক্তি (radiant energy)। সব পদার্থের গঠনে রয়েছে বিভিন্ন প্রকারের মৌলের পরমাণু। পরমাণুর গঠনে রয়েছে কিছু মূল কণিকা। এ সব কণিকার আহিয়ার, শন্যক্তবরণ ও পরমাণুতে এদের বিন্যাস প্রকরণ সম্বন্ধে পরীক্ষা-নিরীকালর তথ্যের যুক্তিনির্ভর উপস্থাপন ওণগত রসায়নের অন্তর্ভক্ত। পরমাণুর সাথে রয়েছে বিকীর্থ শক্তির সম্পর্ক। আবার সব প্রকার বিকীর্থ শক্তি হলো তড়িং চুম্বকীয় বিকিরণ রিপু, যার প্রধান উৎস হলো সূর্যের আলো। আলোক শক্তি-তরঙ্গ নিজে অদৃশ্য; কিছু যে আলোক-অংশ অন্য বস্তুকে দৃশ্যমান করে তাকে দৃশ্যমান আলো বলে। এছাড়াও অনেক অদৃশ্য আলো আছে; যেমন IR রিশ্মি, UV রিশ্মি, রেডিভ-টেলিভিশন তরঙ্গ। এ সব তরঙ্গ রিশ্মি আমরা বিভিন্নভাবে ব্যবহার করি। অনুহুগভাবে, বিভিন্ন প্রমাণু ভাপশক্তি ছারা উত্থীপিত হয়ে বিভিন্ন প্রকার বর্ণান্ধি সৃষ্টি করে, যা থেকে ঐ সব মৌলের শনাক্তকরণ সম্ভব হয়। অজৈব আয়নিক যৌগের মূলক শনাক্তকরণ ও সমযোজী জৈব যৌগের মিশ্রণ থেকে উপাদান যৌগ পৃথকীকরণ, বিশ্লোধন প্রভঙ্তি রাসায়নিক ও ভৌত-প্রক্রিন্যা ওণগত রসায়নের অন্তর্ভক।

অধ্যামের প্রধান শক্ষসমূহ (Key Words): আগফা কণা, তড়িৎ চ্ছকীয় বিকিরণ, ফ্রিক্যুয়েপি ও হাটজ (Hz), ফোটন, প্লাছের সমীকরণ, রেখা বর্ণালি, জিম্যান প্রভাব, বামার-রিডবার্গ সমীকরণ, কোয়ান্টাম সংখ্যা, অরবিটাল, নোড (node), পাউলির বর্জন নীতি, আউফরাউ নীতি, হুতের নীতি, IR রশ্মি, Laser রশ্মি, UV রশ্মি, UV-Fluorescent ink, MRI পরীক্ষা, Flame Test, প্রাব্যতা ও দ্রাব্যতার ওণফল, সম-আয়ন প্রভাব, কেলাসন, উর্ধ্বপাতন, পাতন, দ্রাবক নিকাশন, কোমাটের্য়াকি



# শিবনদদ : এ অধ্যায় পাঠ শেষে শিকার্থীরা---

- পরমাণুর রাদারফোর্ড মডেল ও বোর মডেলের তুলনা করতে পারবে।
- ২. কোয়ান্টাম সংখ্যাতিত্তিক বিভিন্ন উপস্তর ও ইলেব্রন ধারণ ক্ষমতা ব্যাখ্যা করতে পারবে।
- কোল্ডাম উপত্রের শক্তিক্রম ও আকৃতি বর্ণনা করতে পারবে।
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- বোর পরমাণু মভেল অনুসারে 11-পরমাণুর বর্ণালির ব্যাখ্যা করতে পারবে।

# **Hsc Chemistry 2nd Paper 2014**

**Rachel S Tattersall** 

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