

Physical Chemistry

INSTRUCTION: Select the correct answer for each of the following sections. Mark only one answer for each item by shading the box corresponding to the letter of your choice on the answer sheet provided. STRICTLY NO ERASURES ALLOWED. Use pencil no. 1 only.

- Chemical equilibrium reactions are characterized by
 - $\Delta G > 0$
 - $\Delta G < 0$
 - $\Delta G = 0$
 - $\Delta G = \Delta G^\circ$
- Spontaneous reactions are characterized by
 - $\Delta G > 0$
 - $\Delta G < 0$
 - $\Delta G^\circ = 0$
 - $\Delta G = 0$
- Capillary action is due to:
 - Cohesion
 - Viscosity
 - Gravity
 - Surface tension
- An ideal gas has a compressibility factor of
 - 1
 - 2
 - PV
 - None of the given choices
- The fraction of the total current carried by an ion is known as
 - Ionic mobility
 - Overvoltage
 - Transference number
 - Ionic conductance
- A measure of disorderliness in a system is given by
 - Free energy
 - Enthalpy
 - Heat
 - Entropy
- There is no change in internal energy of an ideal gas during
 - Adiabatic expansion
 - Adiabatic compression
 - Isothermal expansion
 - None of the above
- The efficiency of a heat engine is limited by
 - First law of thermodynamics
 - Second law of thermodynamics
 - Third law of thermodynamics
 - None of the above
- Non-ideal gas behavior is likely to be observed at
 - Low pressure and low temperature
 - High pressure and high temperature
 - High pressure and low temperature
 - Low pressure and high temperature
- In an equilibrium reaction, K_p equals K_c when
 - Reactants and products are all gases
 - Temperature is constant
 - Reactants and products have equal number of moles
 - Reactants and products are all liquids
- Natural chlorine is a mixture of isotopes. Determine its atomic mass if 75.53% of the naturally occurring element is chlorine-35, which has a mass of 34.968. Chlorine-37 has a mass of 36.958.
 - 35.45
 - 35.00
 - 34.55
 - 36.47

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