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0806 PHARMACEUTICAL CHEMISTRY-I Mcq Question Bank

Properties of Calcium carbonate

(a) Decomposes in contact with oxidisable matter, reducing agents, on making alkaline, or standing. 2 H2O2→2 H2O+ O2

(b) It acts as an oxidizing or reducing agent depending upon the chemical environment.

(c) Soluble in water, alcohol and ether.

(d) It is insoluble in water and alcohol and water solubility is increased by the presence of CO2 and also by ammonium salt Answer

Properties of Sodium thiosulphate

(a) It is a fine, white, microcrystalline powder.

(b) It is soluble in water but insoluble in alcohol. Answer

(c) It is odourless and tasteless.

(d) It is insoluble in water and alcohol and water solubility is increased by the presence of CO2 and also by ammonium salt

Properties of Potassium permanganate

(a) colourless monoclinic prisms or as a crystalline powder

(b) It is Strong oxidizing agent. Answer

(c) It occurs as transparent.

(d) It is efflorescence in dry air and deliquescence in moist air.

Properties of Hydrogen peroxide

(a) It has a cool and bitter taste.

(b) It is soluble in water but insoluble in alcohol.

(c) It is freely soluble in water, soluble in glycerine and insoluble in alcohol.

(d) Colourless and odourless liquid with slight acidic taste. Answer

Properties of Calcium carbonate

(a) It is soluble in water.

(b) It is a strong acid & attacks metals forming their hydrochlorides with the evolution of hydrogen gas.

(c) It is insoluble in water and alcohol and water solubility is increased by the presence of CO2 and also by ammonium salt. Answer

(d) At elevated temperatures, nitrous oxide is a powerful oxidizer similar to molecular oxygen

Pharmaceutics Mcq Questions With Answers

Lilian M. Azzopardi

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