

Chemistry – Unit 6 Worksheet 1 – Electrolysis

15. A sealed spring-piston cylinder containing nitrogen gas also contains a thin partition. The pressure of the gas on both sides of the thin partition is 100 kPa. The gas is N_2 on the left and the temperature of the gas increases to 100°C. What is the pressure on the right?

100 kPa (the volume changes but not the pressure)

100 kPa (the volume increases so the pressure decreases)

100 kPa (the volume increases so the pressure increases)

not enough information

100 kPa (the volume is constant)

100 kPa (the pressure is constant)

100 kPa

16. A 100 mL sample of nitrogen gas is collected over water at 25°C and 100 kPa. The gas is collected over water. The pressure of the gas is 100 kPa. The pressure of the water vapor is 3.2 kPa. What is the pressure of the dry gas?

100 kPa (the pressure of the gas is 100 kPa)

100 kPa (the pressure of the gas is 100 kPa)

100 kPa (the pressure of the gas is 100 kPa)

100 kPa (the pressure of the gas is 100 kPa)

17. A sealed cylinder contains a mixture of gases. The pressure of the gas is 100 kPa. The pressure of the water vapor is 3.2 kPa. What is the pressure of the dry gas?

100 kPa (the pressure of the gas is 100 kPa)

100 kPa (the pressure of the gas is 100 kPa)

100 kPa (the pressure of the gas is 100 kPa)

100 kPa

Modeling Chemistry U9 Ws 2 V2 Answers

Arthur James Wells



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The British National Bibliography Arthur James Wells, 1979
Theory, 1977

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