

# Molecular Polarity and Electronegativity: A PHET Online Inquiry lab

Name: \_\_\_\_\_ Section: \_\_\_\_\_ Date: \_\_\_\_\_

**PHET Simulation: Molecular Polarity**

**Directions:**

- Go to the following website: <https://phet.colorado.edu/en/simulation/molecular-polarity>
- Click the word "SIMULATIONS" at the top of the screen and choose Chemistry from the drop down menu.
- Find the simulation called Molecular Polarity and click it to open it.
- Click the play button.

Or go directly to this address in your browser:  
<https://phet.colorado.edu/en/simulation/molecular-polarity>

First, click on the box above the words "Three atoms"

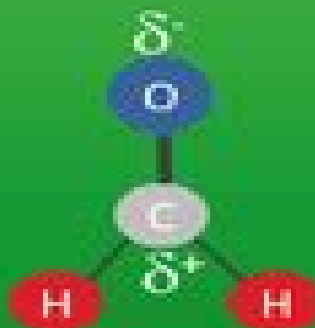
Experiment with the simulator. Try out the controls, click and drag things to move them around, click buttons and try to get an idea of how the simulator works. When you finish, push the reset button  and then answer the questions below.

1) Which atom has the highest electronegativity?

Click the check boxes for bond dipole and partial charges

2) After do you think that the molecule dipole arrow points upward? Consider the bond dipoles.

3) Is the molecule polar? Explain. Consider the partial charges of each atom.



Name: \_\_\_\_\_ Section: \_\_\_\_\_ Date: \_\_\_\_\_

**Molecular Polarity Questions**

1) How does the molecular dipole arrow in the simulator tell you if molecule is polar or nonpolar and tell you the partial positive and negative charges are in the molecule?

2) How do the bond dipole arrows relate to whether or not a molecule is polar?

3) When you turn on the electric field below questions 4 and 5, why does the molecule interact with the electric field in that way it did?

4) When you turn on the electric field below question 5, why does the molecule interact with the electric field in that way that it did?

A CHEMISTRY INQUIRY LAB  
WORKED ANSWER KEY INCLUDED  
GRADING GUIDE INCLUDED

# Phet Molecular Structure And Polarity Lab Answers

**Daniela Niemeyer**



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