

### QUESTION

Determine the boiling point of ethylene glycol to water solution of different composition

### STEP 1

1. Determine the boiling point of pure water: The boiling point of pure water is  $100^{\circ}\text{C}$  at standard atmospheric pressure.

### STEP 2

2. Determine the boiling point of pure ethylene glycol: The boiling point of pure ethylene glycol is  $197.3^{\circ}\text{C}$  at standard atmospheric pressure.

# Determine The Boiling Point Of Ethylene Glycol Water Solution Of Different Composition

**Therald Moeller**



## **Determine The Boiling Point Of Ethylene Glycol Water Solution Of Different Composition:**

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provides illustrations and problems covering the field of physical chemistry The material has been arranged into illustrations that are solved and supplemented by problems thus enabling readers to determine the extent to which they have mastered each subject Most of the illustrations and problems were taken from original papers to which reference is made The English edition of this book has been translated from the manuscript of the 2nd Czech edition It has been changed slightly in some places and enlarged on in others on the basis of further experience gained in teaching physical chemistry at the Institute of Chemical Technology in Prague The book begins with illustrations and problems on the atomic structure and the fundamentals of quantum mechanics Subsequent chapters cover the kinetic theory of ideal gas fundamentals of thermodynamics states of matter phase equilibrium chemical equilibrium and third law of thermodynamics electrochemistry reaction kinetics surface phenomena and colloidal systems and molecular structure and physical properties

**Laboratory Manual for Principles of General Chemistry** Jo Allan Beran, 2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures

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*Official Gazette of the United States Patent and Trademark Office* United States. Patent and Trademark Office, 1999

**Chemistry** John A. Olmsted, Gregory M. Williams, Robert Charles Burk, 2016-01-14 Olmsted Burk is an introductory general chemistry text designed specifically with Canadian professors and students in mind A reorganized Table of Contents and inclusion of SI units IUPAC standards and Canadian content designed to engage and motivate readers distinguish this text from many of the current text offerings It more accurately reflects the curriculum of most Canadian institutions Instructors will find the text sufficiently rigorous while it engages and retains student interest through its accessible language and clear problem solving program without an excess of material that makes most text appear daunting and redundant

**A Textbook of Physical Chemistry** Arthur Adamson, 2012-12-02 A Textbook of Physical Chemistry Second

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